



Experimental results of HFO/HCFO refrigerants in a laboratory scale HTHP with up to 150 °C supply temperature

C. Arpagaus, S.S. Bertsch

NTB Buchs

- **Introduction to high temperature heat pumps (HTHP)**
- **Suitable HFOs and HCFOs for HTHPs**
- **System design of the laboratory scale HTHP at NTB Buchs**
- **Experimental results with R1336mzz(Z), R1233zd(E) and R1224yd(Z)**
- **Conclusions**



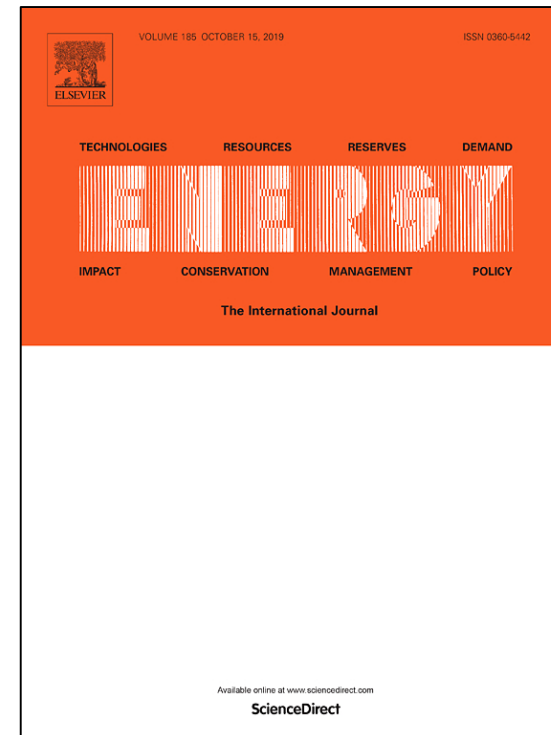
Introduction to high temperature heat pumps (HTHP)

Review Papers

Arpagaus C., Bless F., Schiffmann J.,
Bertsch S.S.: [Multi-temperature heat pumps: A literature review](#), International Journal of Refrigeration, 2016, 69, 437–465.

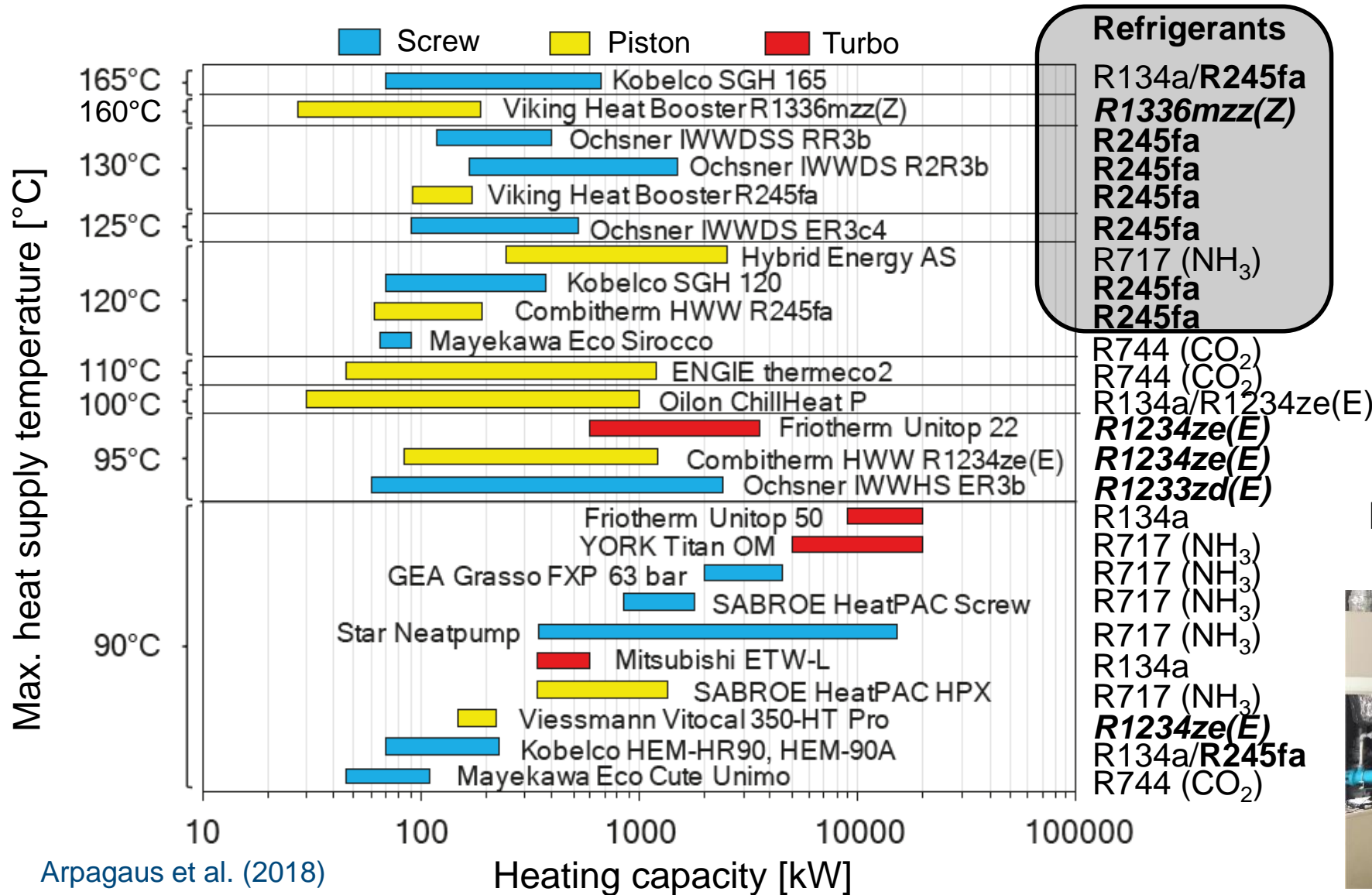


Arpagaus C., Bless F., Uhlmann M., Schiffmann J., Bertsch S.S.: [Review - High temperature heat pumps: Market overview, state of the art, research status, refrigerants, and application potentials](#), Energy, 2018, 152, 985-1010



> 26 industrial HTHP products with heat supply temperature ≥ 90 °C available

R245fa is predominantly used in industrial HTHP ... but has a high GWP of 858



Arpagaus et al. (2018)

Research gaps in High Temperature Heat Pumps

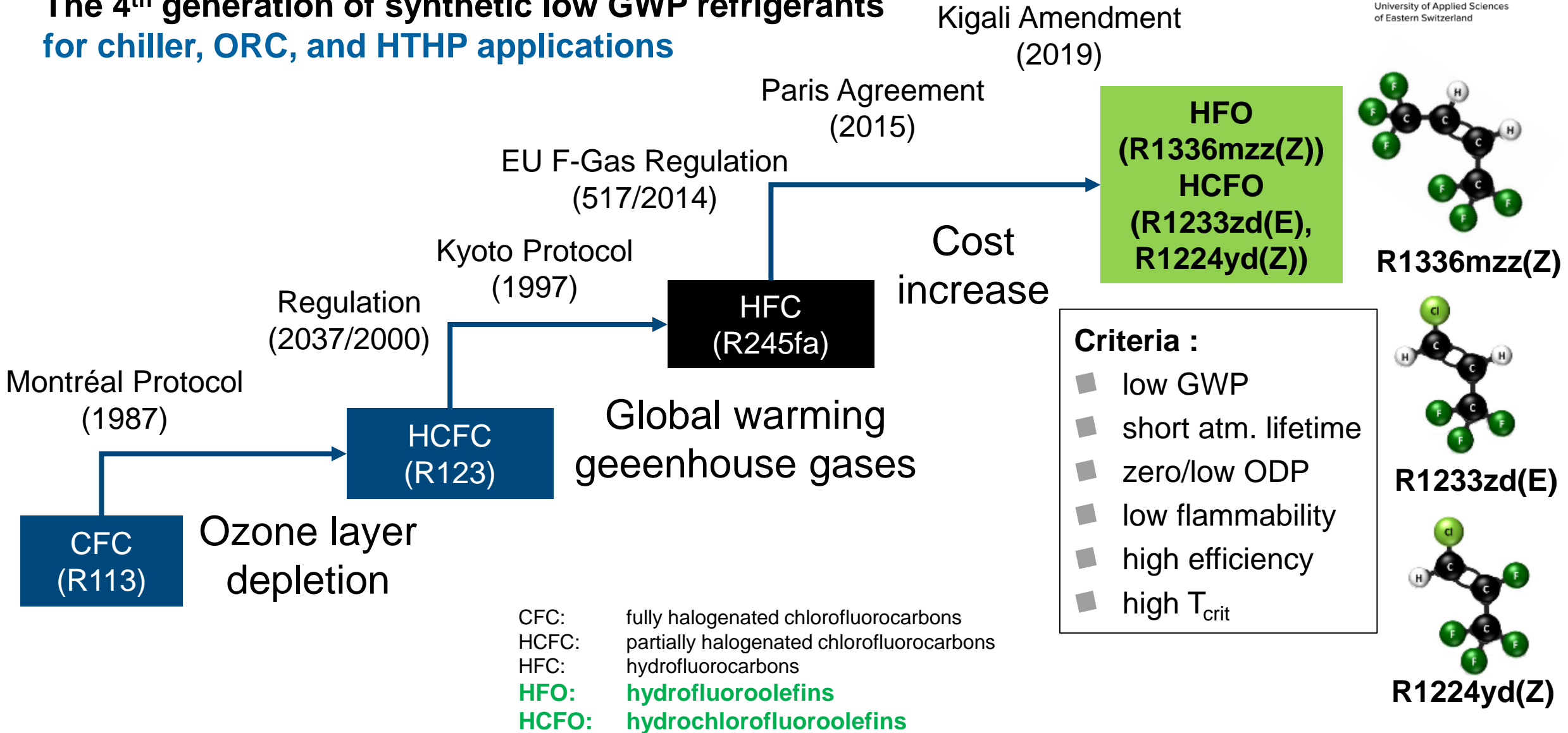


Focus

- Application of natural refrigerants, such as hydrocarbons (R600, R601), CO₂ or water
- Extending heat source/sink to higher temperatures
- Improving heat pump efficiency (COP) (e.g. by multi-stage cycles, oil-free compressors)
- Development of temperature-resistant components (e.g. valves, compressors)
- New control strategies for higher temperatures
- Scale-up of functional models to industrial scale
- **Testing of new environmentally friendly synthetic refrigerants for HTHPs (e.g. HFOs and HCFOs)**

Suitable HFOs and HCFOs for HTHPs

The 4th generation of synthetic low GWP refrigerants for chiller, ORC, and HTHP applications

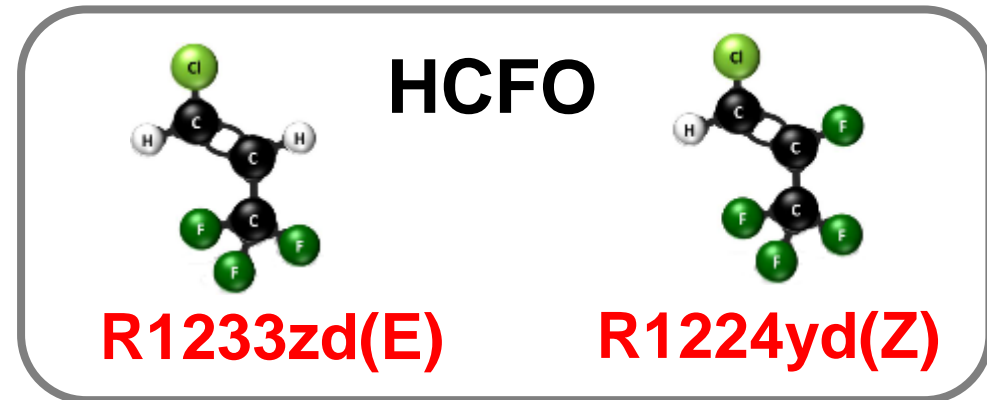
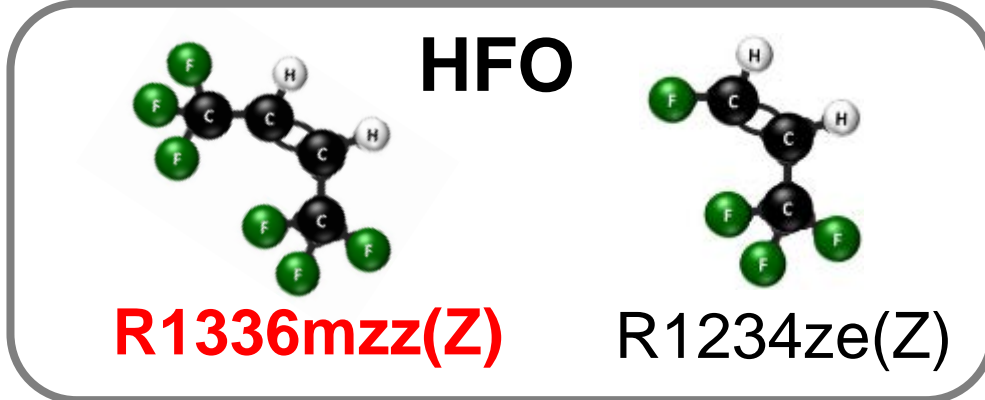


Properties of suitable HFO and HCFO refrigerants for HTHPs

Tested
in this
study



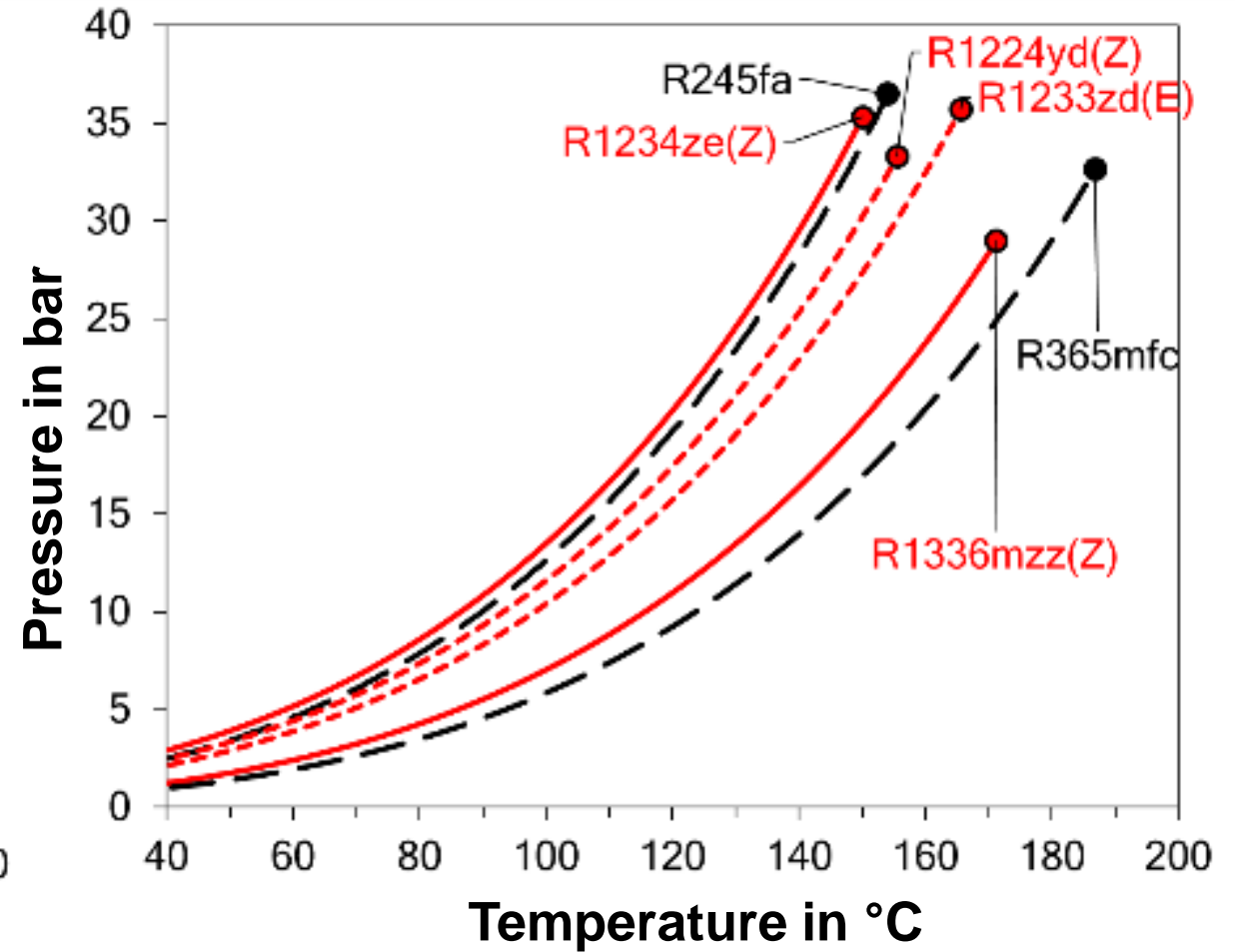
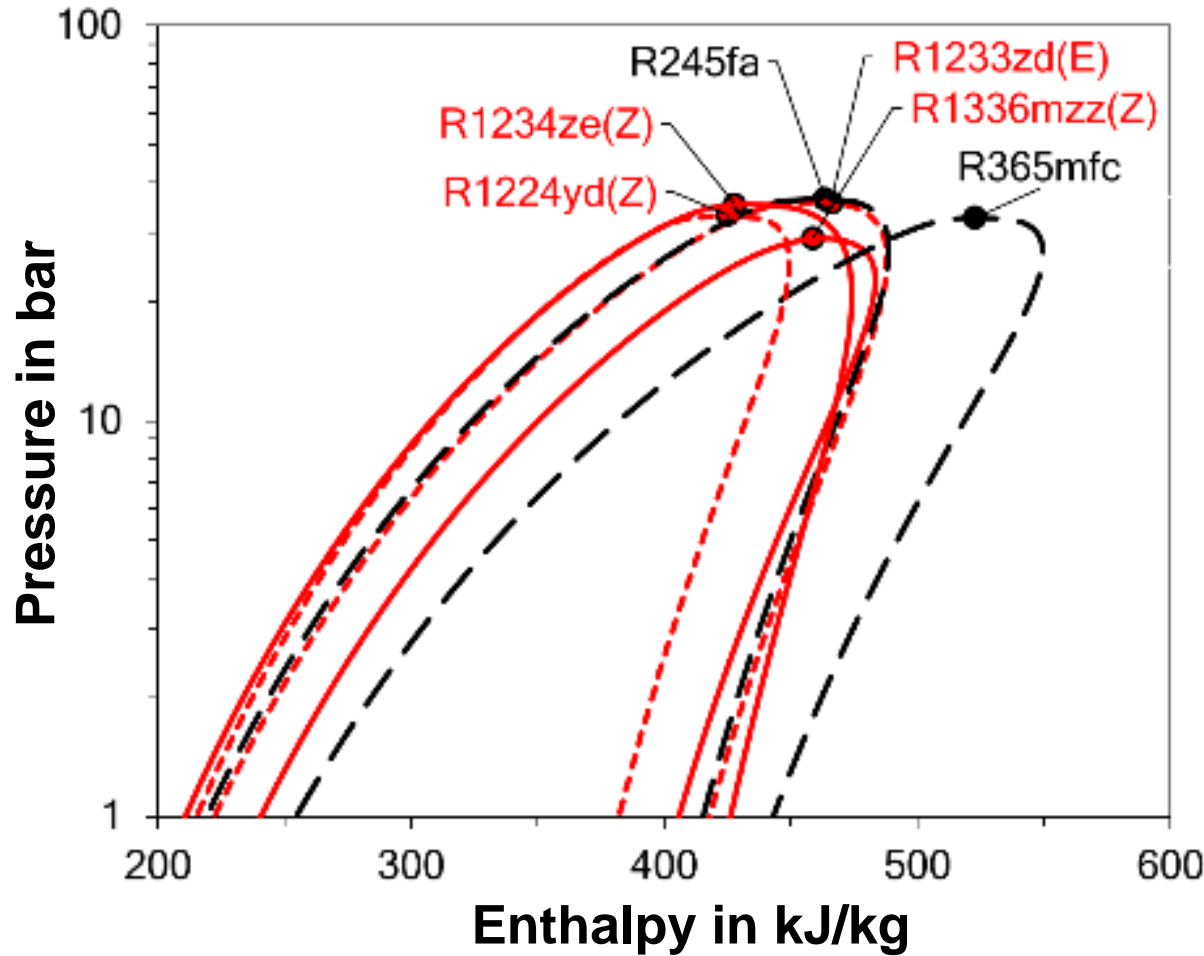
Refrigerant	Brand (manufacturer)	T _{crit} [°C]	p _{crit} [bar]	ODP [-]	GWP ₁₀₀ [-]	Lifetime [days]	SG	NBP [°C]
R1336mzz(Z)	Opteon™ MZ (Chemours)	171.3	29.0	0	2 ^a	22 ^a	A1	33.4
R1234ze(Z) ^b	Not yet available	150.1	35.3	0	<1 ^a	10 ^a , 18 ^b	A2L	9.8
R1233zd(E)	Solstice®zd (Honeywell) Forane®HTS 1233zd (ARKEMA)	165.6	35.7	0.00034 ^d , 0.00030 ^e	1 ^a , <5 ^e	~14 ^f , 26 ^a , 36 ^e , 40.4 ^d	A1	18.0
R1224yd(Z)	AMOLEA®1224yd (AGC Chemicals)	155.5	33.3	0.00023 ^c	0.88 ^c	20 ^c	A1	14.0
R365mfc	Solkane®365mfc (Solvay)	186.9	32.7	0	804 ^a	8.7 years ^a	A2	40.2
R245fa	Genetron® 245fa (Honeywell)	154.0	36.5	0	858 ^a	7.7 years ^a	B1	14.9



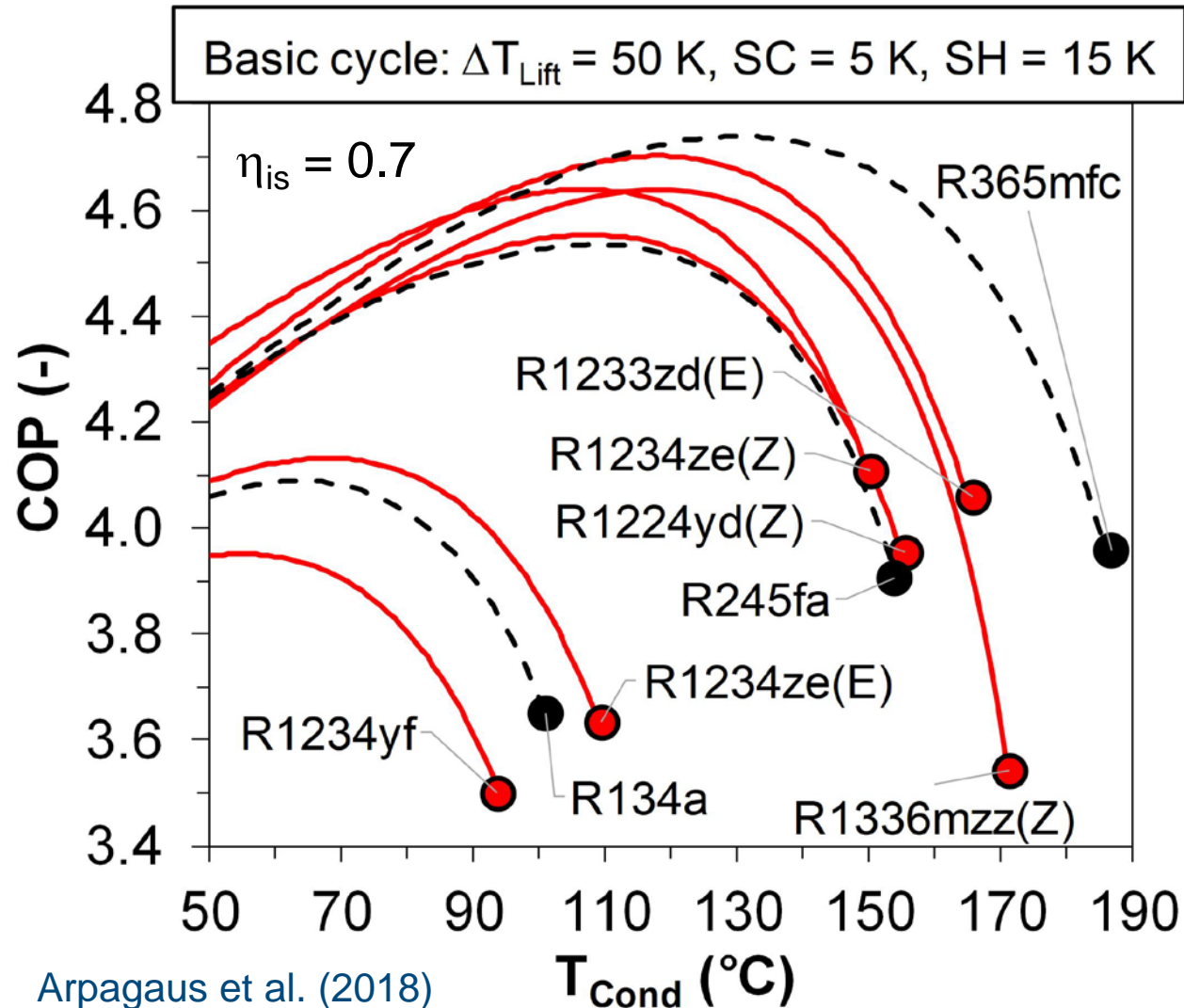
References:

T_{crit} and p_{crit} (EES F-Chart Software, V10.643, 2019), ODP basis R11=1.0 (UNEP, 2017), GWP₁₀₀ (100-year time horizon, CO₂=1.0), SG: Safety group classification (ASHRAE 34, 2016), ^aMyhre et al. (2013, IPCC 5th assessment report), ^bFukuda et al. (2014), ^cTokuhashi et al. (2018), ^dPatten and Wuebbles (2010), ^eSulbaek Andersen et al. (2018) (3D global model), ^fAndersen et al. (2015)

T-s, p-T diagrams of selected HFO and HCFO refrigerants

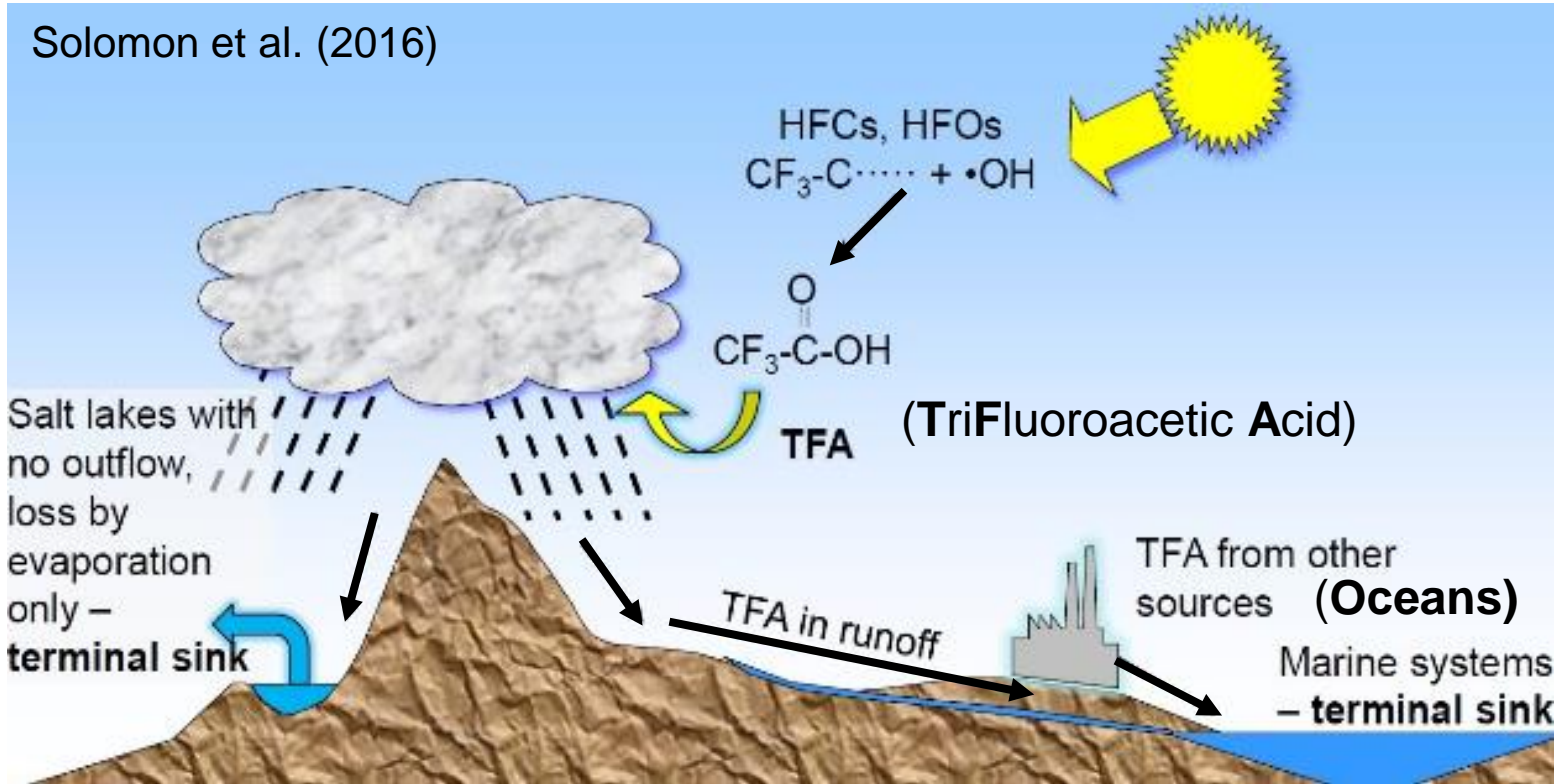


Simulated COP of selected HFO and HCFO refrigerants in a basic HP cycle



- Simulated COP rise to an optimum and decrease with the narrowing of the 2-phase region up to T_{crit}
- Optimal COP at about 30 K below the critical temperature
- R365mfc offers highest COP, followed by R1233zd(E) and R1336mzz(Z)
- R1234ze(Z) and R1224yd(Z) comparable to R245fa
- R1234yf and R1234ze(E) similar to R134a

Environmental fate of TFA (trifluoroacetic acid, $\text{CF}_3\text{C}(\text{O})\text{OH}$)



- 268 million tons TFA are present in the oceans, i.e. non-anthropogenic
- 200 ng/L average TFA concentration in oceans (Frank et al., 2002)

Upper range scenario:

- Total TFA yield from 1990 up to 2050 = 20.625 million tons TFA (Solomon et al., 2016)

↓ : 1.34×10^{21} L oceans water

Negligible risk for aquatic organisms and human health

Total additional contribution of TFA from HFCs and HFOs to the oceans is 15.3 ng/L* or <7.5% of the naturally present TFA

Atmospheric degradation products of HFOs and molar yields of TFA formation (Trifluoroacetic acid, $\text{CF}_3\text{C}(\text{O})\text{OH}$)

	Refrigerant	Formula	Final degradation products	Molar yields of TFA $\text{CF}_3\text{C}(\text{O})\text{OH}$
HFO	R1234yf	$\text{CF}_3\text{-CF=CH}_2$	$\text{CF}_3\text{C}(\text{O})\text{OH}$, CO_2 , HF	100%
	R1234ze(E)	$\text{E-CF}_3\text{-CH=CHF}$	CO_2 , $\text{HC}(\text{O})\text{OH}$, HF	<10%, 0%
	R1336mzz(Z)	$\text{Z-CF}_3\text{-CH=CHCF}_3$	CO_2, HF	<20%^a
HCFO	R1233zd(E)	$\text{E-CF}_3\text{-CH=CHCl}$	CO_2 , HF, HCl	~ 2% ^b
	R1224yd(Z)	$\text{Z-CF}_3\text{-CF=CHCl}$	<i>similar structure like R1234yf degrading to $\text{CF}_3\text{C}(\text{O})\text{F}$ and hydrolyzing to TFA</i>	
HFC	R365mfc	$\text{CF}_3\text{-CH}_2\text{-CF}_2\text{-CH}_3$	CO_2 , HF	<10%
	R245fa	$\text{CHF}_2\text{-CH}_2\text{-CF}_3$	CO_2 , HF	<10%

TFA formation yield depends on HFO refrigerant

Risk of TFA formation for R1336mzz(Z) and R1233zd(E) is considered to be close to negligible

Products:

$\text{CF}_3\text{C}(\text{O})\text{OH}$ trifluoroacetic acid (TFA)
 $\text{HC}(\text{O})\text{OH}$ formic acid
 CO_2 carbon dioxide
 HCl hydrochloric acid
 HF hydrofluoric acid

References:

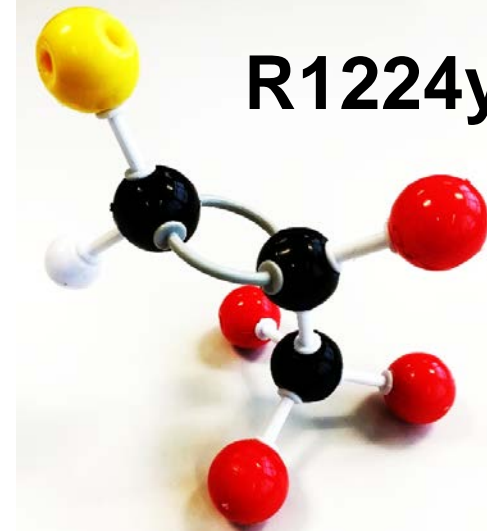
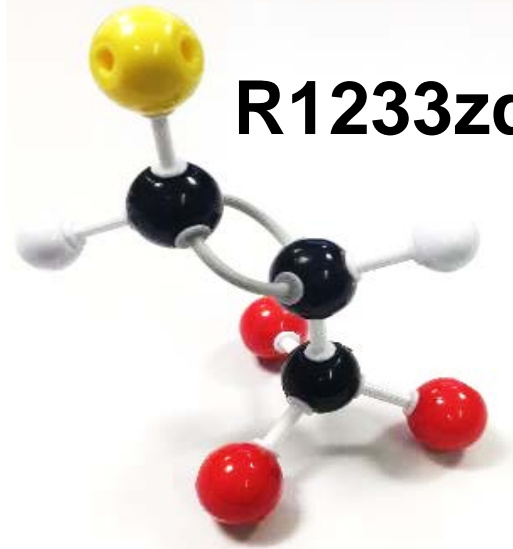
Norwegian Environment Agency (2017), WMO (2018), Wallington et al. (2014), Juhasz & Kontomaris (2018), EFCTC (2019), ^aHenne et. al. (2012), ^bSulbaek Andersen et al. (2008, 2012, 2018), Inoue et al. (2008), ECETOC (2004), Chen et al. (1997)

Goals of this study



Objectives:

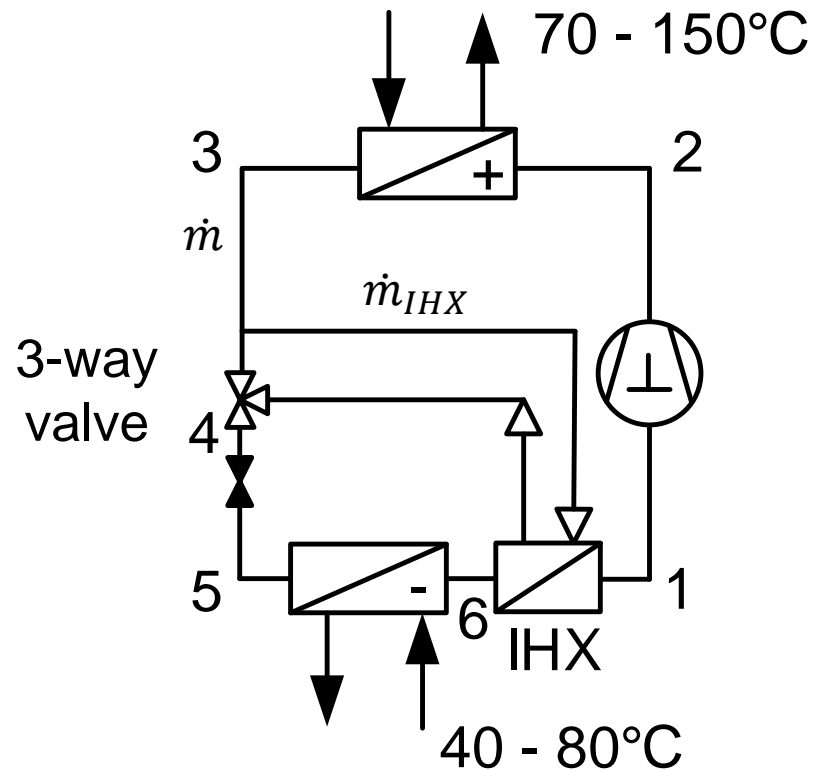
- **Performance evaluation of R1336mzz(Z) (Opteon™MZ, Chemours), R1233zd(E) (Solstice®zd, Honeywell), and R1224yd(Z) (AMOLEA®1224yd, AGC Chemicals) in a laboratory HTHP (drop-in test).**



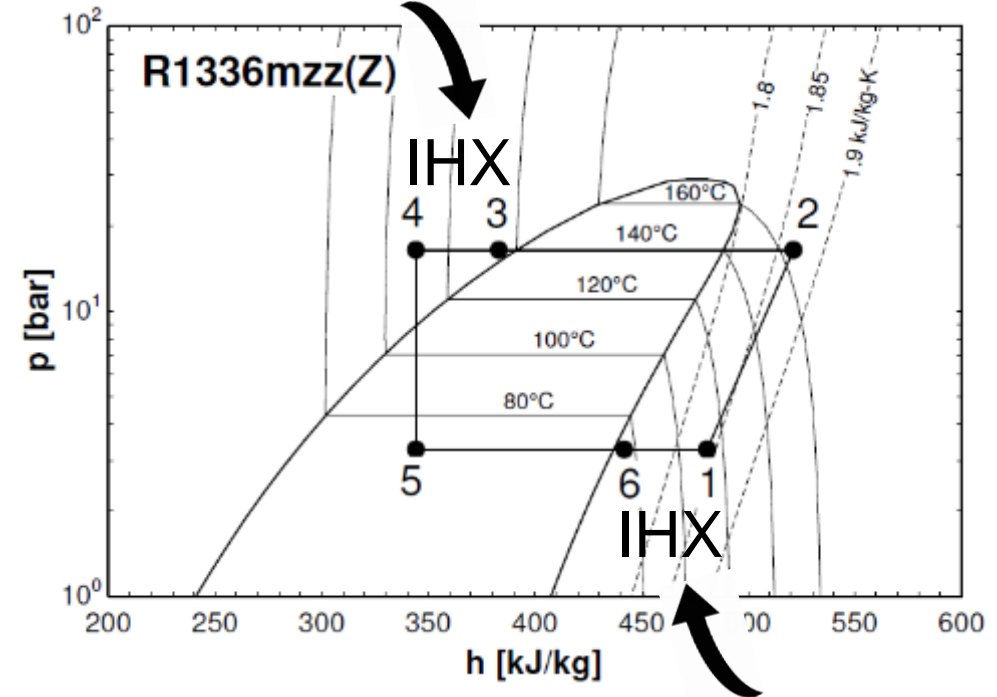
**+ POE oil
(SE 170)**

System design of the laboratory scale HTHP at NTB Buchs

1-stage cycle with internal heat exchanger (IHX) and adjustable 3-way valve

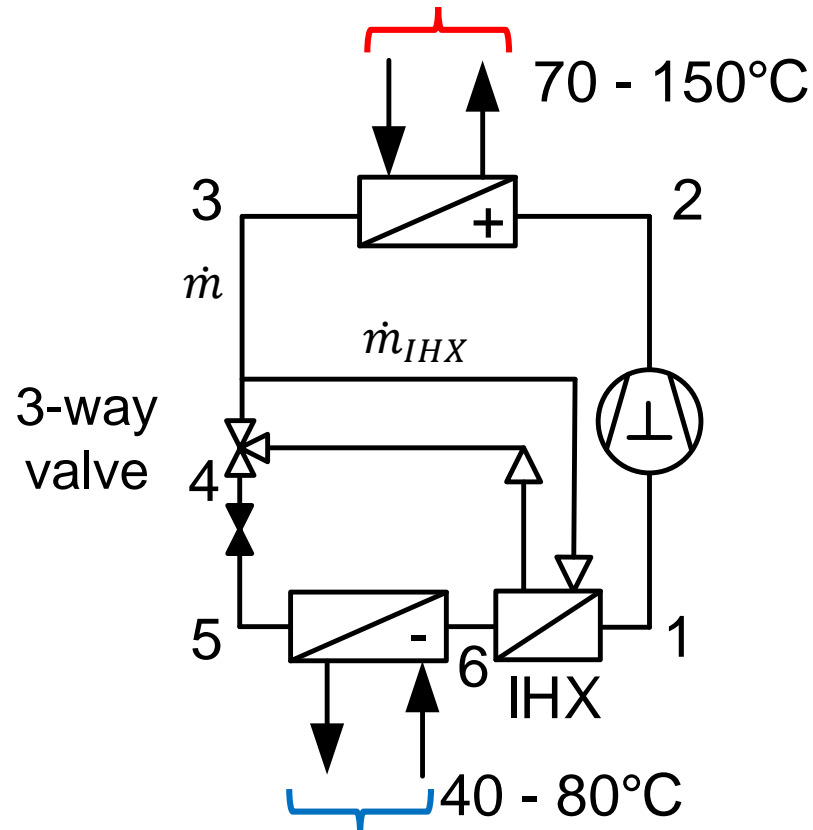


$IHX: \dot{m}_{IHX} / \dot{m} = 0\% \rightarrow 100\%$
 (Opening degree 3-way-valve)



Reference conditions and variation range (water/water heat pump)

$\Delta T_{\text{Sink}} = 5 \text{ K (Ref) to } 25 \text{ K (Temperature glide)}$



$\Delta T_{\text{Source}} = 3 \text{ K (constant)}$

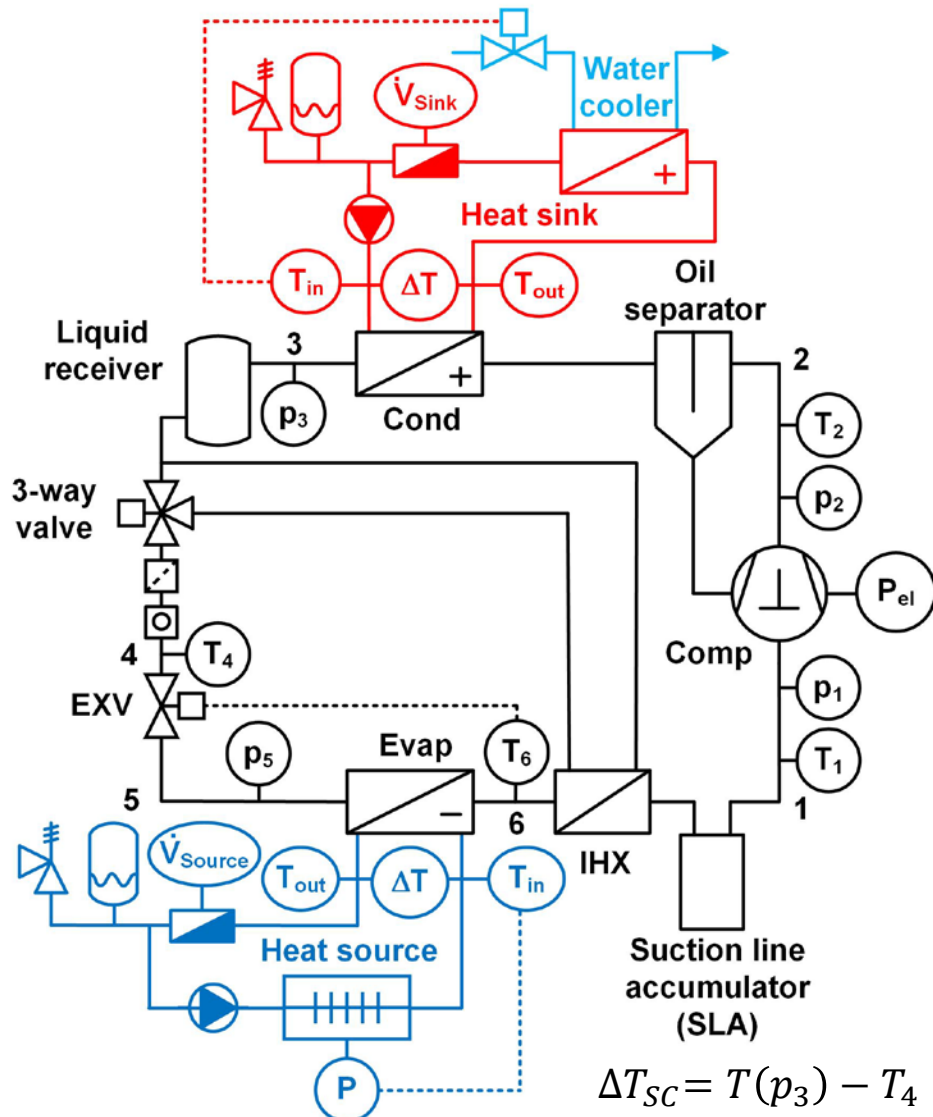
	Reference point (Ref)	Variation range
$T_{\text{Sink,out}}$	$110 \pm 1^\circ\text{C}$	70 to 150°C
$T_{\text{Source,in}}$	$60 \pm 1^\circ\text{C}$	40 to 80°C
ΔT_{Lift}	50 K	30 to 70 K
ΔT_{Sink}	$5.0 \pm 0.1 \text{ K}$	5 to 25 K
ΔT_{Source}	$3.0 \pm 0.1 \text{ K}$	constant
f_{Komp}	50 Hz	constant
IHX (Opening angle of 3-way-valve)	100%	0 to 100%
	$IHX: \dot{m}_{IHX}/\dot{m} = 0\% \rightarrow 100\%$	

Superheating after evaporator:

$$\Delta T_{\text{SH}} = T_6 - T(p_{\text{Evap}}) = 5 \text{ K}$$

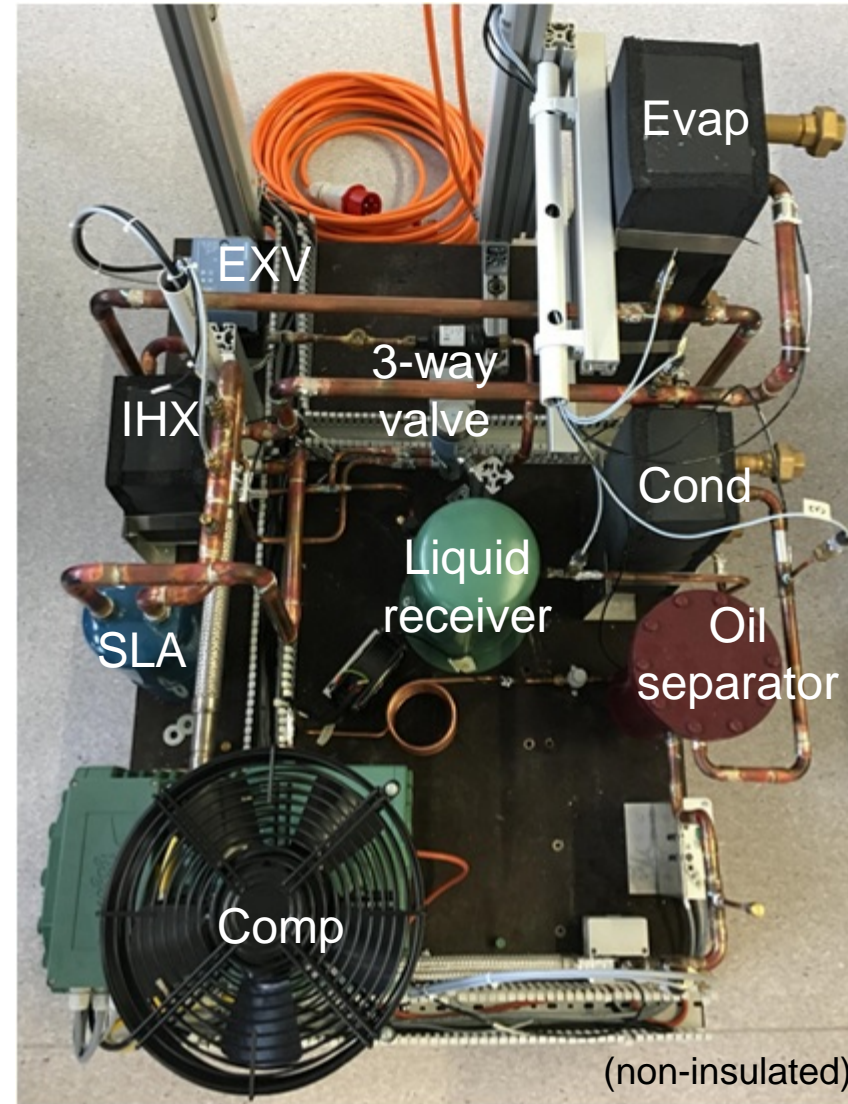
IHX generates additional superheating

Experimental set-up and schematics of the laboratory HTHP



$$\Delta T_{SC} = T(p_3) - T_4$$

$$\Delta T_{SH} = T(p_5) - T_6$$



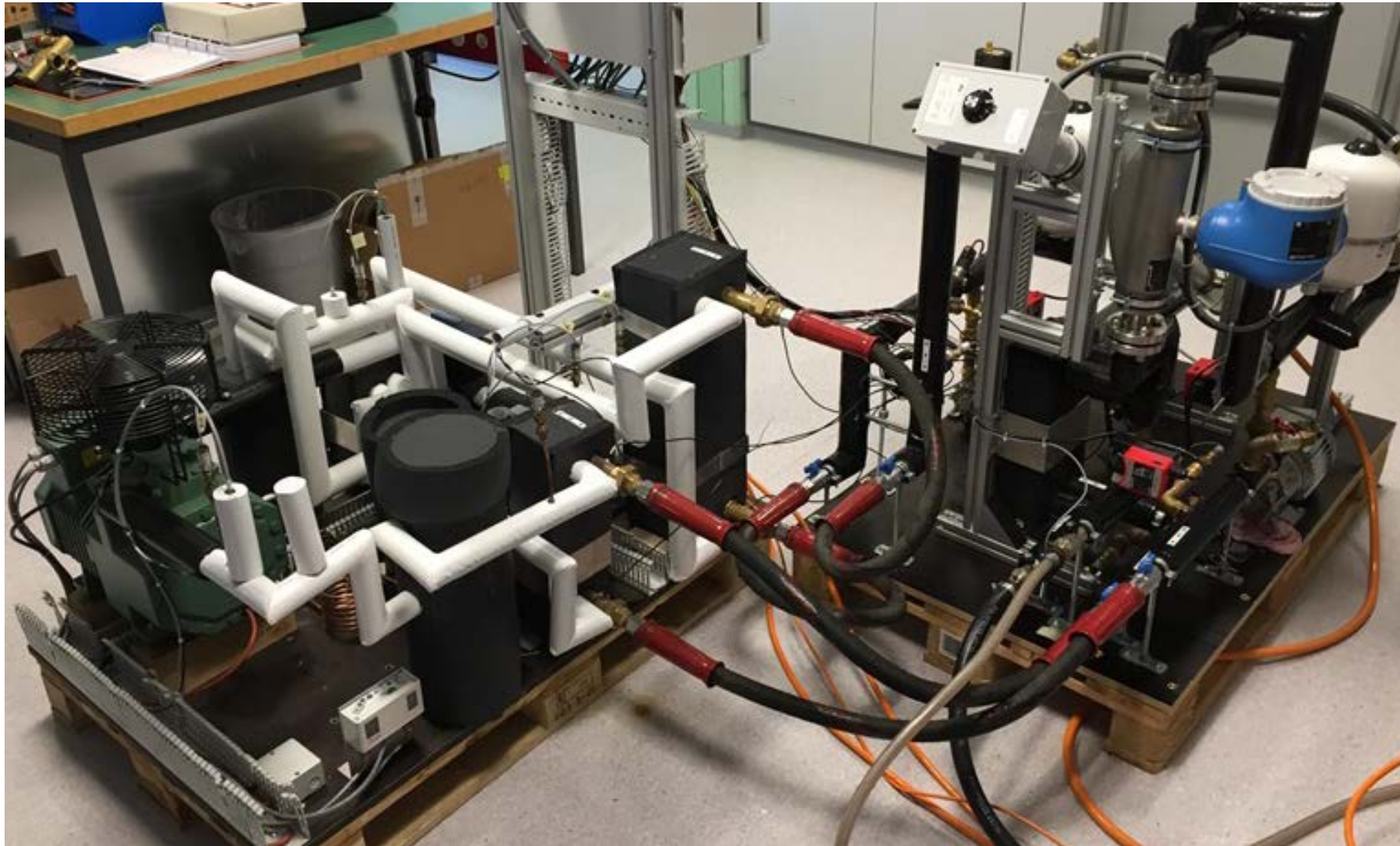
(non-insulated)

POE oil SE 170
 kinematic viscosity
 at 40 °C: 173 cSt
 at 100 °C: 17,6 cSt

Variable-speed
 semi-hermetic
 piston compressor
 Bitzer, 2DES-3Y
 New Ecoline

Motor switch-off
 temperature
 ~110 °C

Laboratory HTHP with hydraulic loops for heat source and sink



Sensors and measurement uncertainties

Measured parameters	Sensor type	Uncertainties
Pressures	$p_{1...6}$ Piezoelectric, 0 to 50 bar, max. 120°C	max. 1.5% of full scale reading
Temperatures	$T_{1...6}$ Thermocouples, type K, class 1	± 1.5 K
Heat sink temp difference	ΔT_{Sink} Thermocouples, type K, class 1	± 0.1 K
Compressor power	P_{Comp} Power transmitter, 0 to 15 kW	0.2 % of measuring range + 0.1 % measured value
Heat sink mass flow (water)	\dot{m}_{H_2O} Coriolis, 0 to 1'300 kg/h, max. 180 °C	± 0.05 %

$$COP = \frac{\dot{Q}_{Sink}}{P_{Comp}} = \frac{\dot{m}_{H_2O} \cdot c_{p,H_2O}(T) \cdot \Delta T_{Sink}}{P_{Comp}}$$

$$COP_{Carnot} = \frac{T_{Sink,out}}{T_{Sink,out} - T_{Source,in}}$$

2nd Law efficiency:

$$\eta_{2nd} = \frac{COP_H}{COP_{Carnot}}$$

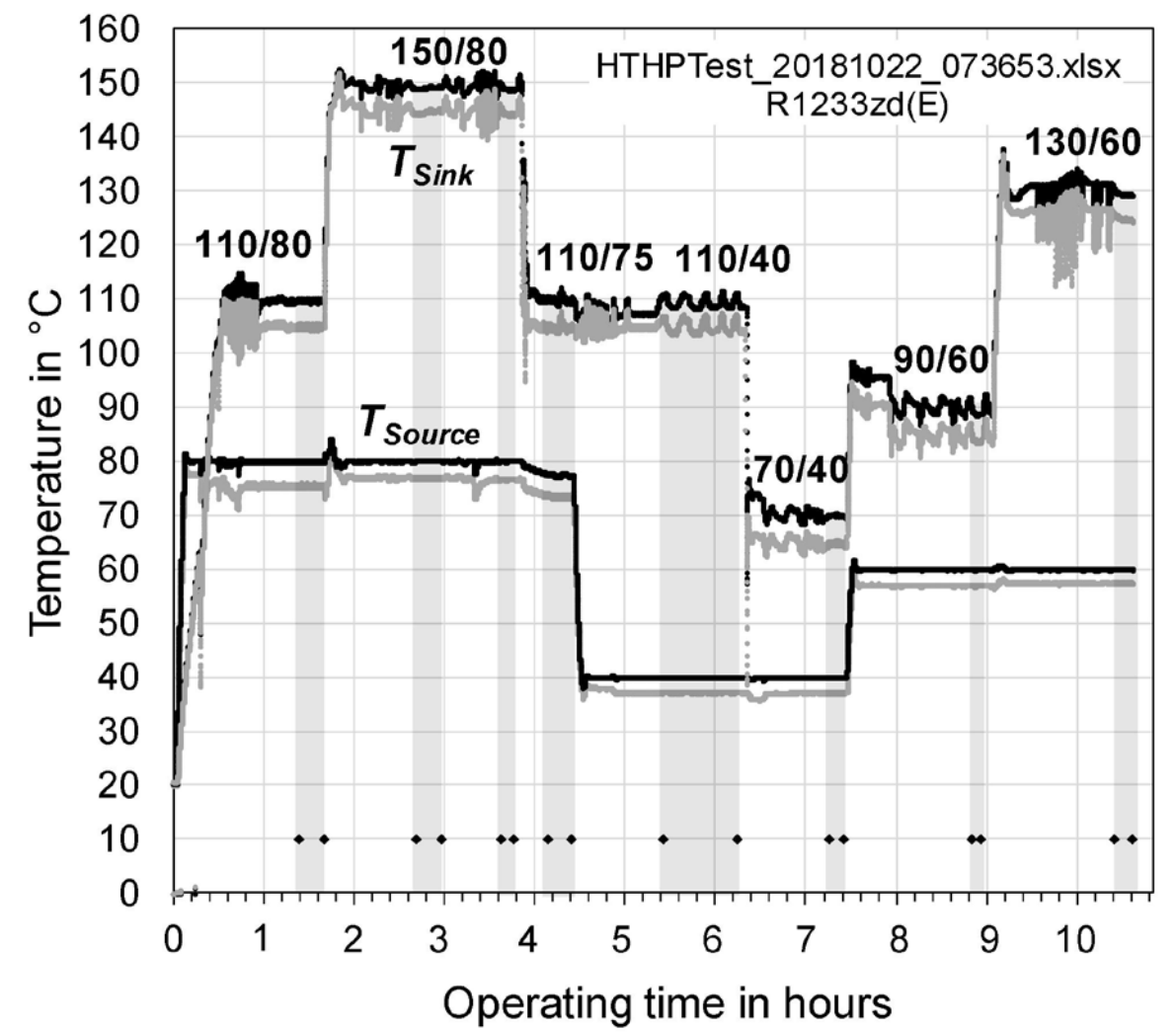
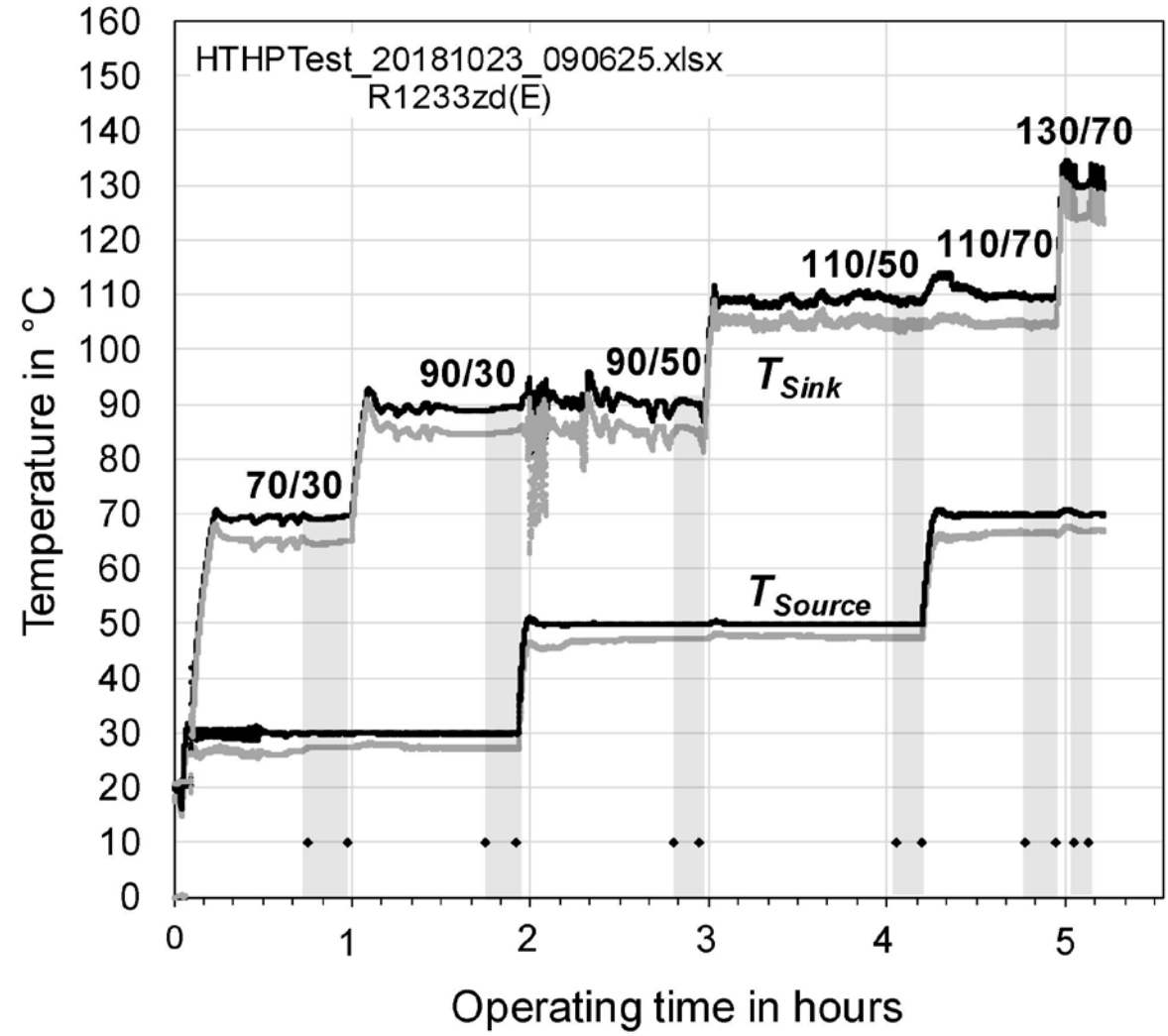
Error propagation according to RSS method (Root Sum Squares):

$$\Delta COP = \sqrt{\left(\frac{\partial COP}{\partial \dot{m}_{H_2O}} \cdot \Delta \dot{m}_{H_2O}\right)^2 + \underbrace{\left(\frac{\partial COP}{\partial c_{p,H_2O}(T)} \cdot \Delta c_{p,H_2O}(T)\right)^2}_{\sim 0} + \left(\frac{\partial COP}{\partial \Delta T_{Sink}} \cdot \Delta(\Delta T_{Sink})\right)^2 + \left(\frac{\partial COP}{\partial P_{Comp}} \cdot \Delta P_{Comp}\right)^2}$$

Average uncertainty	R1336mzz(Z)	R1233zd(E)
ΔCOP	± 0.21 (4.2%)	± 0.21 (4.1%)
$\Delta \dot{Q}_{Sink}$	± 0.14 kW (3.7%)	± 0.22 kW (3.8%)
ΔP_{Comp}	± 0.031 kW (2.6%)	± 0.032 kW (1.7%)

$\Delta T_{\text{Sink}} \quad 5.0 \pm 0.1 \text{ K}$
 $\Delta T_{\text{Source}} \quad 3.0 \pm 0.1 \text{ K}$

Temperature profiles of experimental runs (at least 5 min stable conditions)



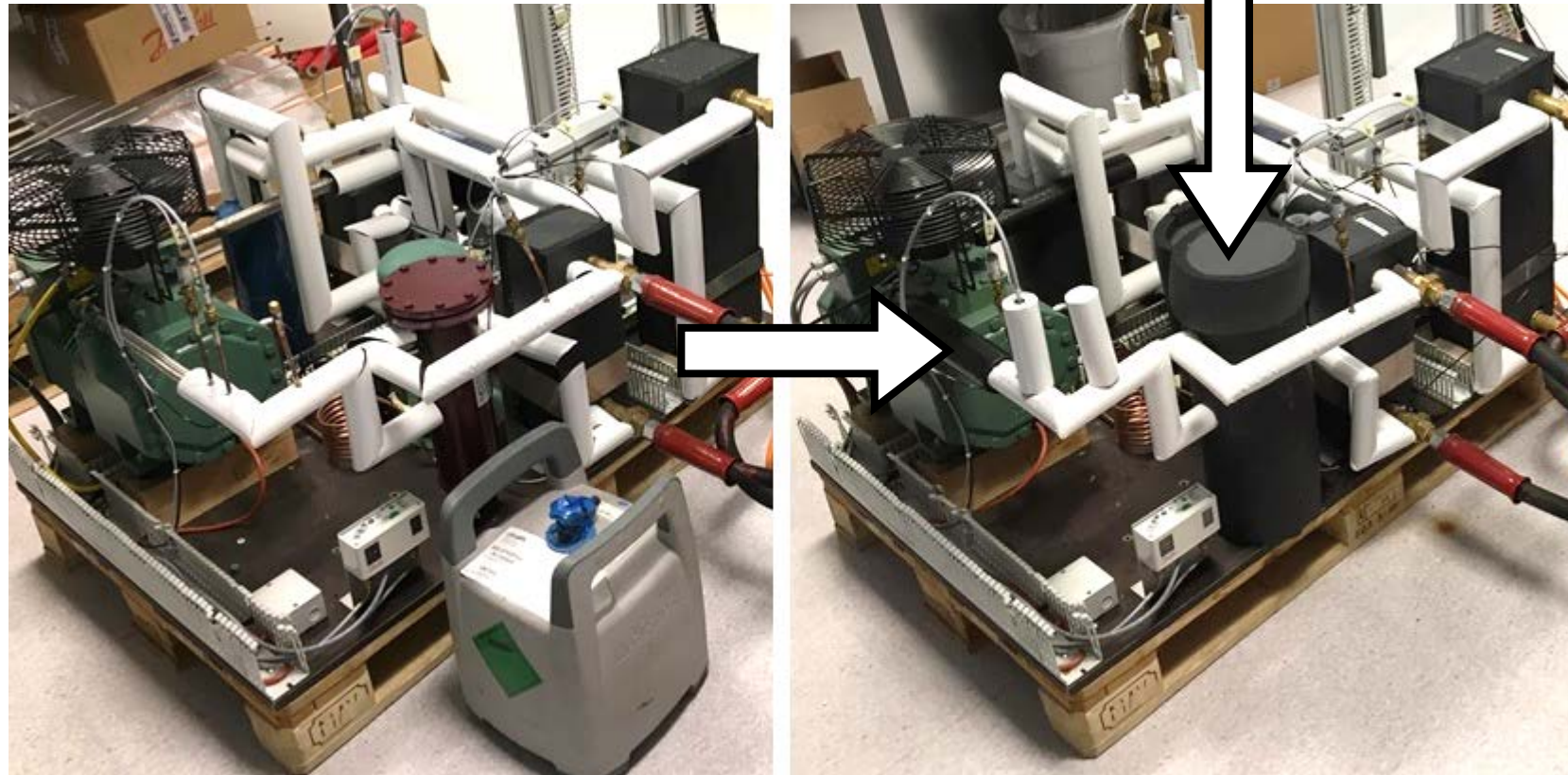
Experimental results with R1336mzz(Z), R1233zd(E) and R1224yd(Z)

COP improvement with better insulation of oil separator, liquid receiver, and suction line accumulator with Armaflex®HT insulation



Temperature resistance: up to 150°C
 Thermal conductivity (0°C): 0.038 W/m

**COP improvement Δ
 with better insulation**



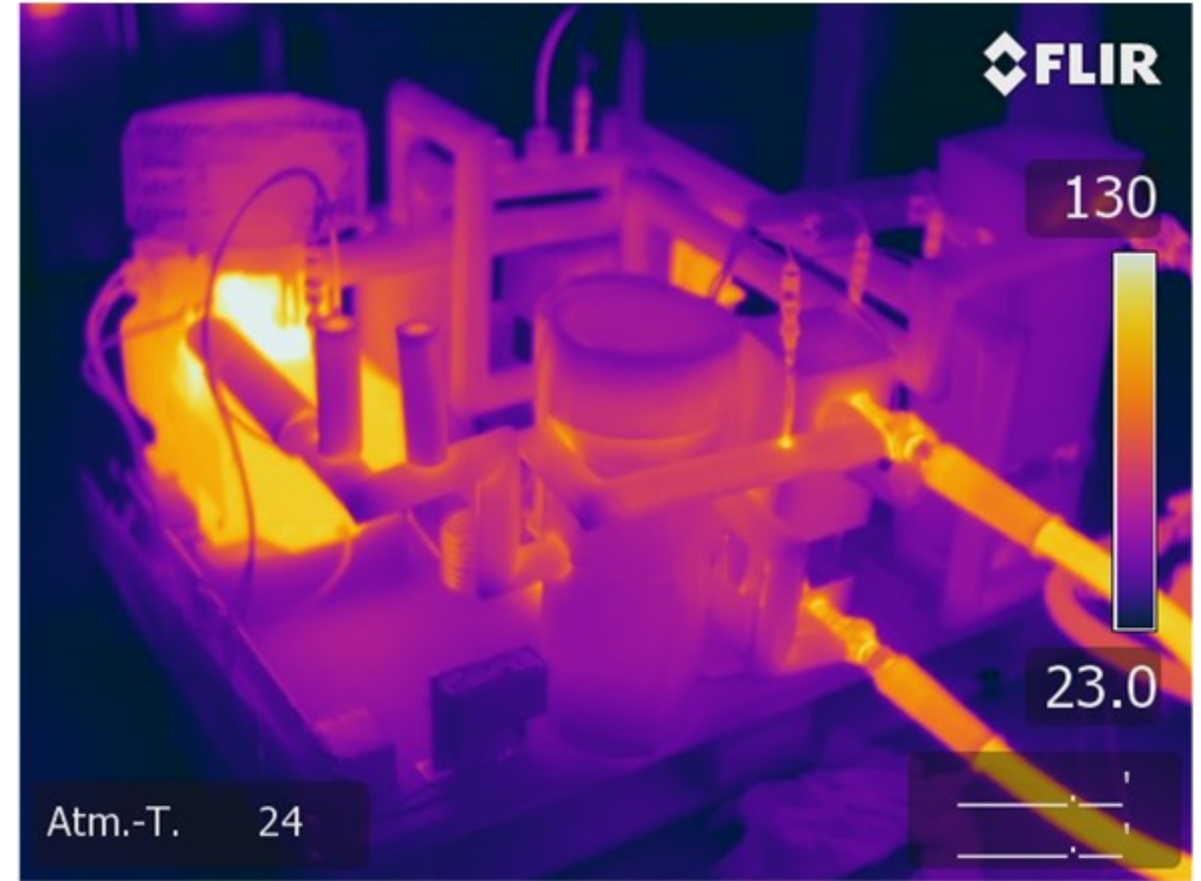
$T_{Source,in} / T_{Sink,out}$ (ΔT_{Lift})	COP (before ¹⁾)	COP (after ²⁾)	Δ
40/90 (50)	2.58	3.14	+22%
60/110 (50)	2.78	3.09	+11%
80/130 (50)	2.67	3.10	+16%

(1-stage cycle with 100% IHX)

¹⁾ Arpagaus et al. (2018), 17th Int. Refrig. Air Cond. Conf., Purdue, July 9-12, 2018.

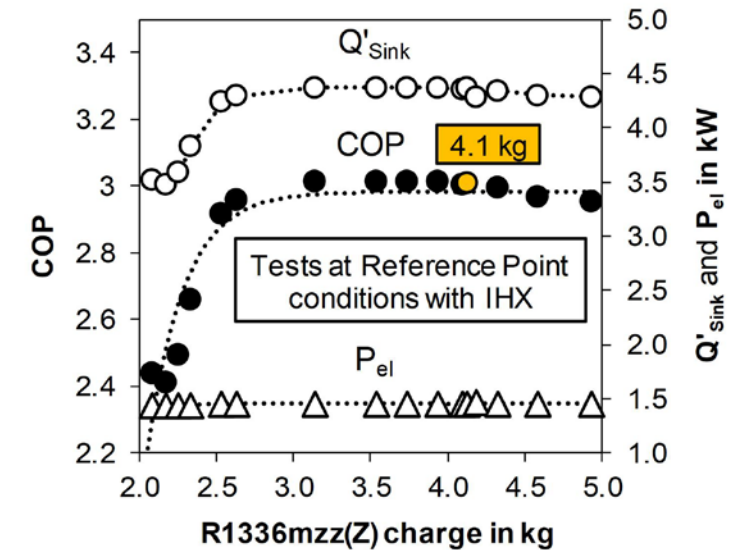
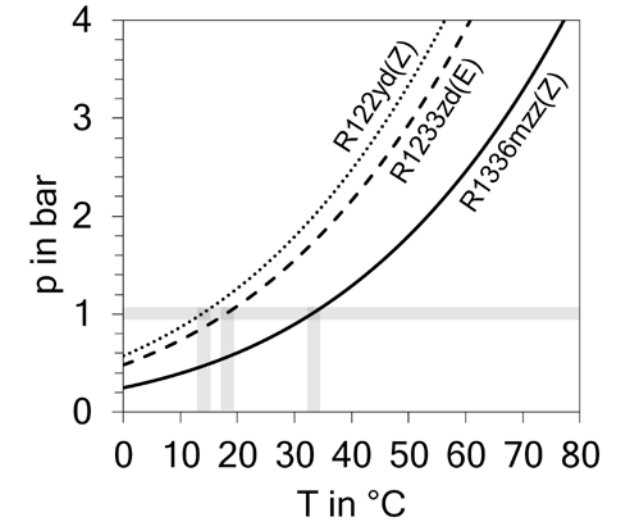
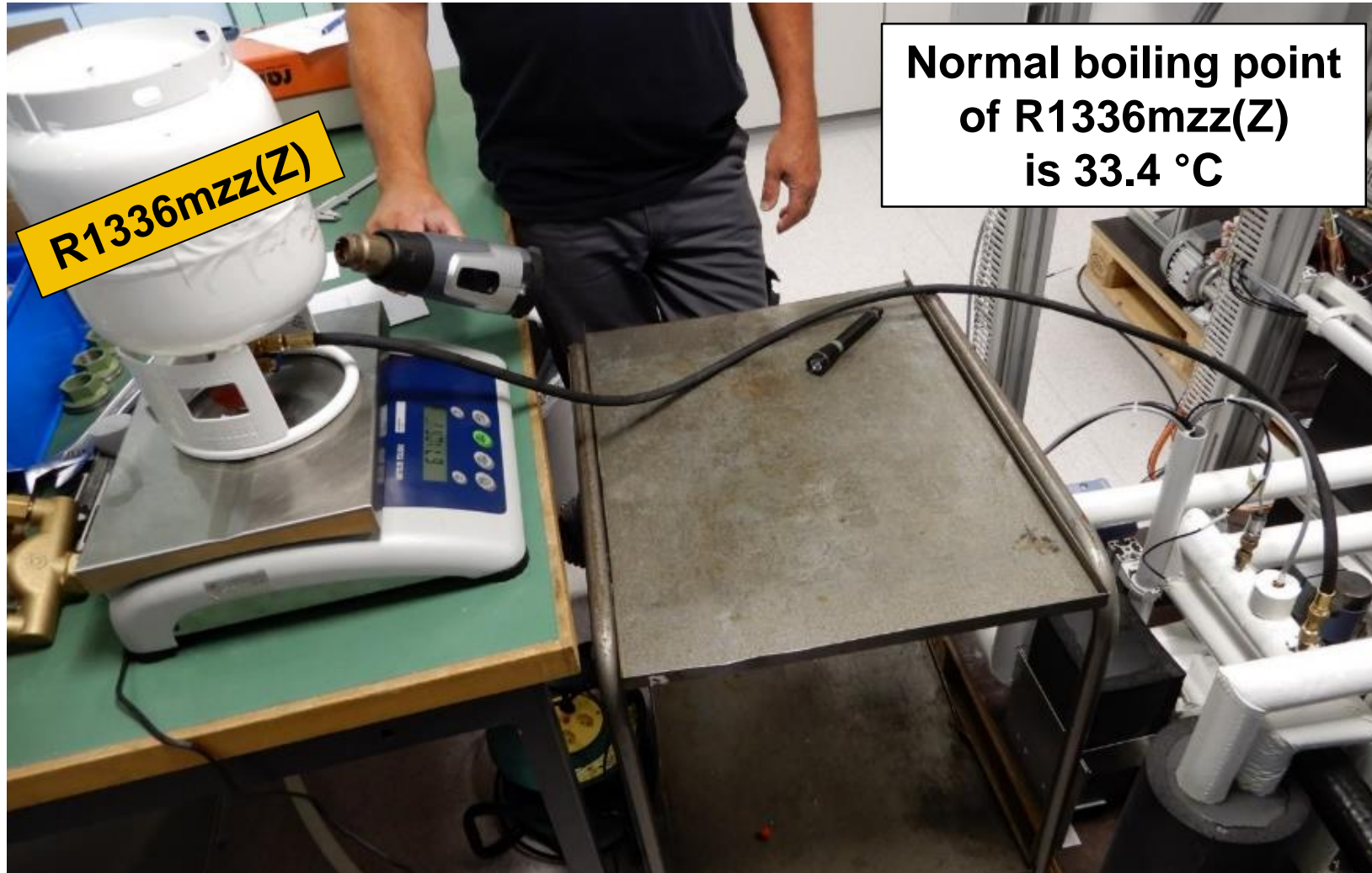
²⁾ Arpagaus et al. (2018), DKV-Tagung 2018, Aachen, November 21-23, 2018.

Infrared camera image for hot spot identification

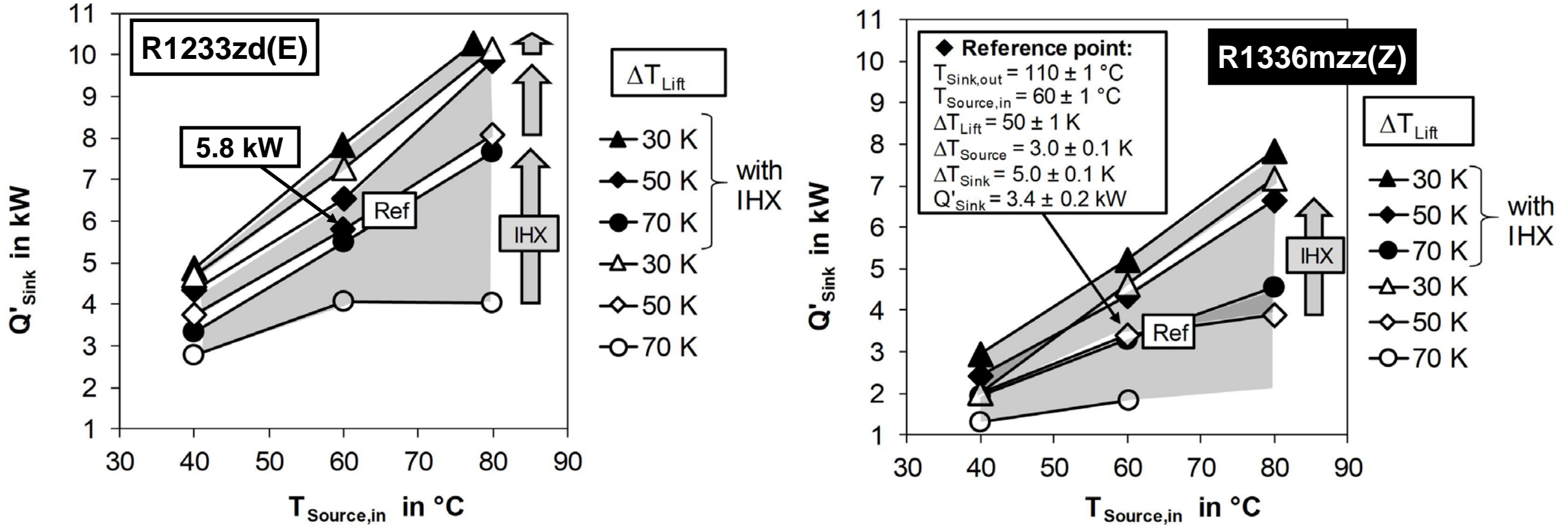


- Heat losses of about 21 ± 7 % estimated from energy balance (major heat losses at the compressor)
- There is still potential for optimization in insulation and possibilities for increasing efficiency

Refrigerant filling procedure with heating-up of the refrigerant cylinder



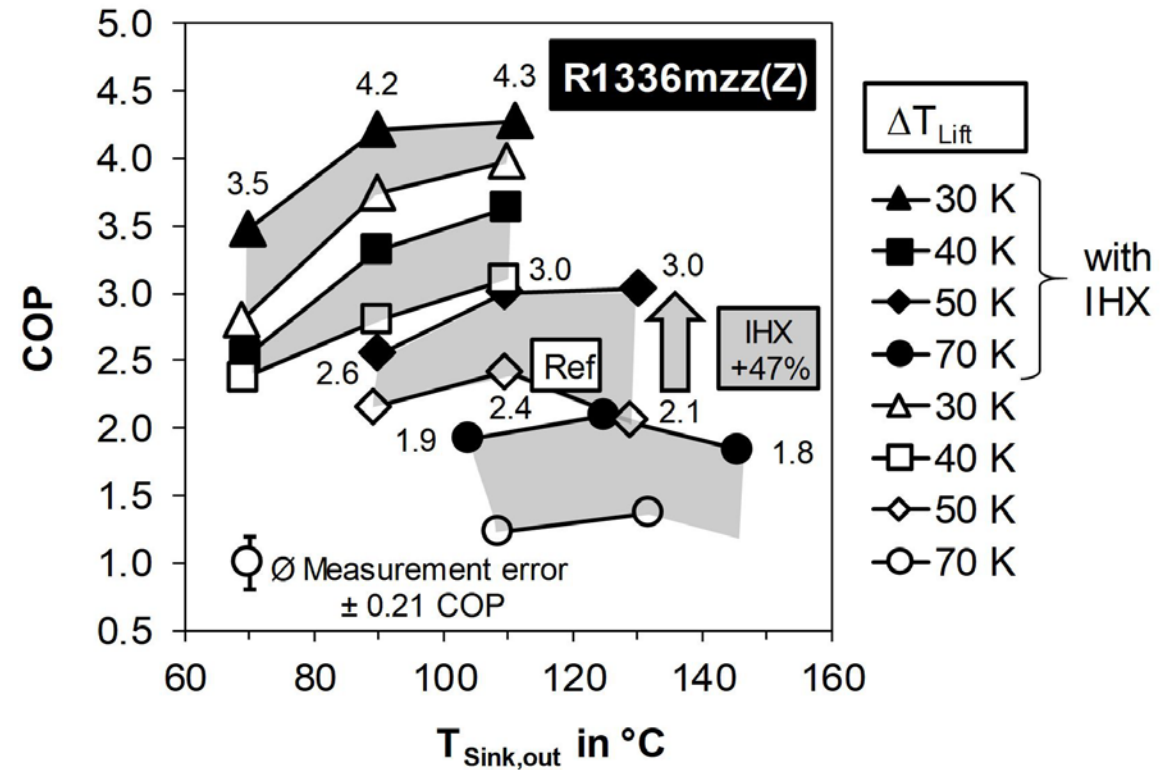
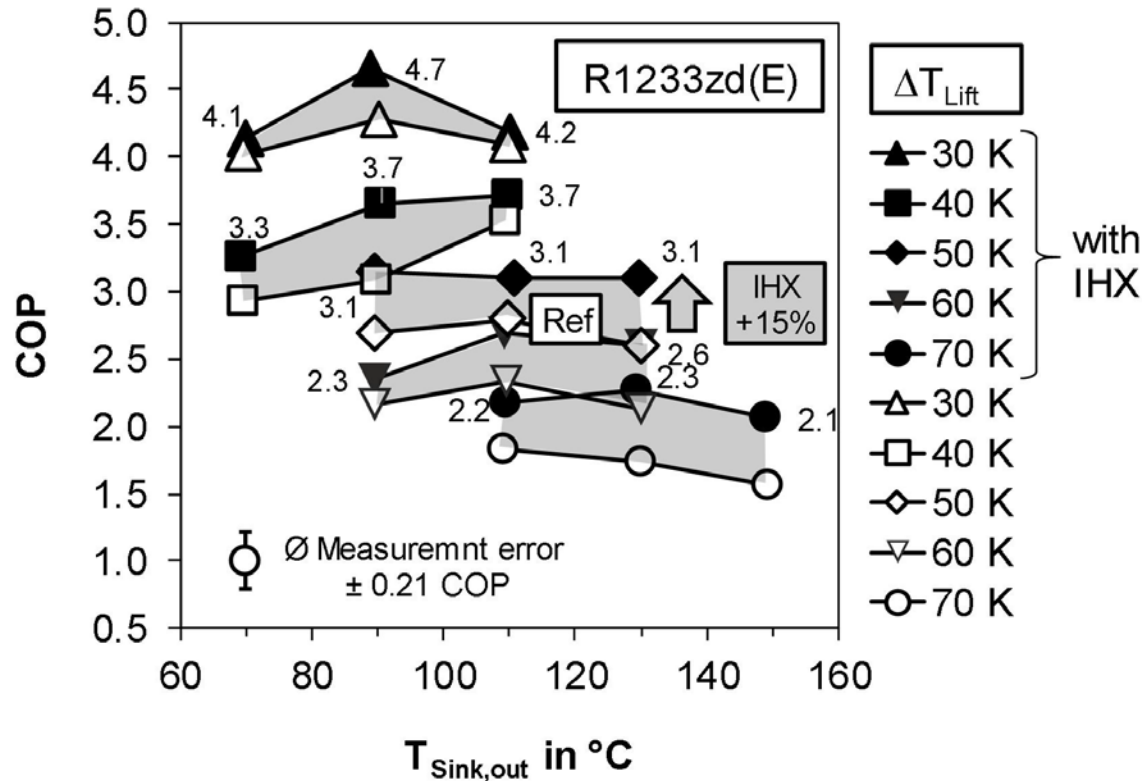
Comparison of heating capacity with the basic and IHX cycle



- R1233zd(E) provides 46 to 76% higher heating capacity than R1336mzz(Z) (e.g. 5.8 vs. 3.4 kW at W60/W110)
- R1336mzz(Z) would require a larger compressor swept volume to achieve similar heating capacities

Operating maps of efficiency with basic and IHX cycle

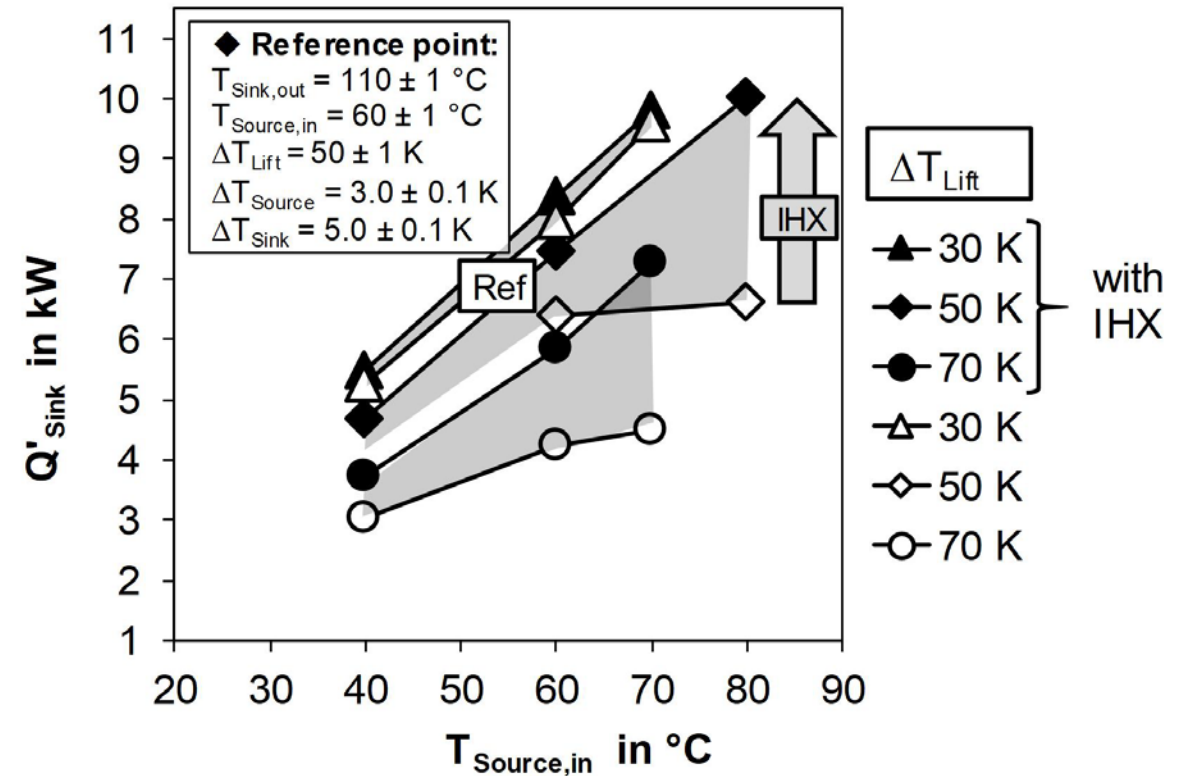
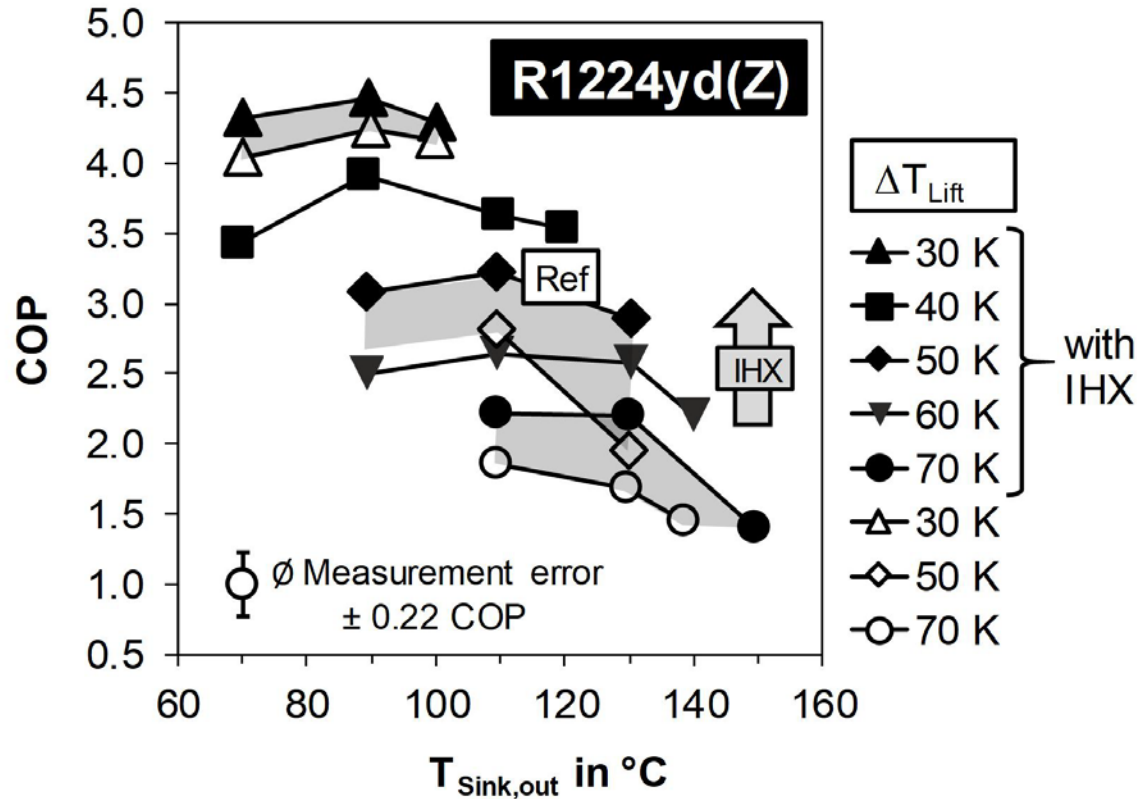
COP vs. $T_{\text{Sink,out}}$ at different temperature lifts ($\Delta T_{\text{Lift}} = T_{\text{Sink,out}} - T_{\text{Source,in}}$) from 30 to 70 K



- As expected, COP increase with smaller ΔT_{Lift} and higher $T_{\text{Sink,out}}$ according to Carnot relationship
- IHX integration provides significant COP increase (superheat achieved in IHX, higher evaporation temperature)
- Max. heat sink temperature tested was 150 °C with a COP of 2.1 for R1233zd(E) and 1.8 for R1336mzz(Z)

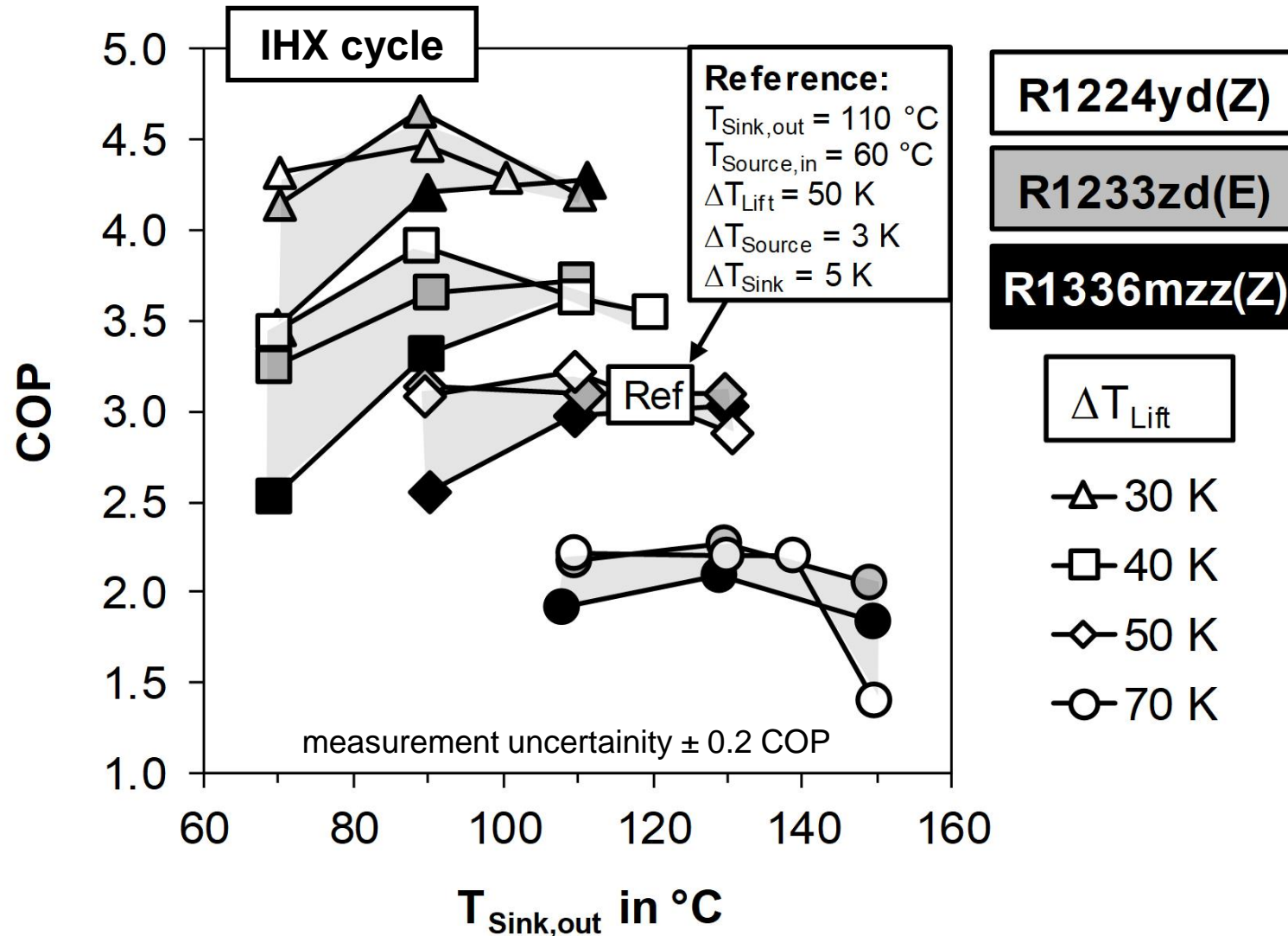
Operating maps of efficiency with basic and IHX cycle

COP vs. $T_{\text{Sink,out}}$ at different temperature lifts ($\Delta T_{\text{Lift}} = T_{\text{Sink,out}} - T_{\text{Source,in}}$) from 30 to 70 K



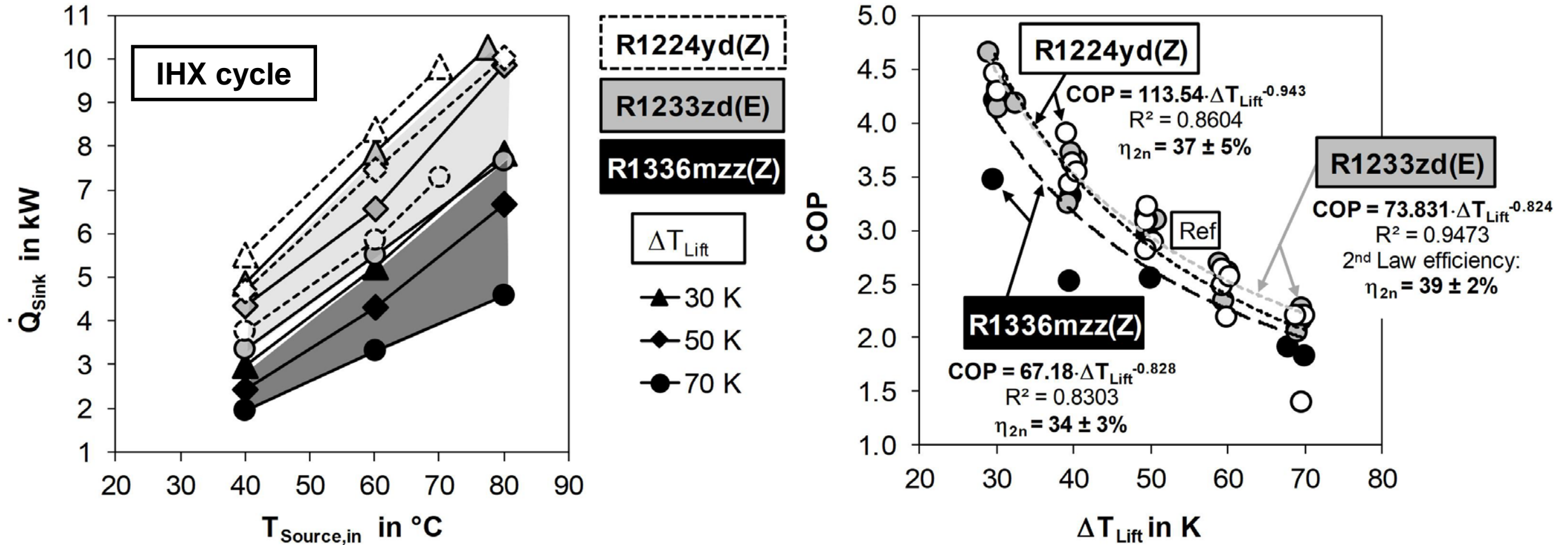
- IHX increases the COP about 14% compared to a basic cycle at Ref conditions
- COP of 3.2 reached at W60/W110 (50 K temperature lift, with IHX)
- 10 kW heating capacity reached at W80/W130 with IHX cycle

Comparison of efficiency maps



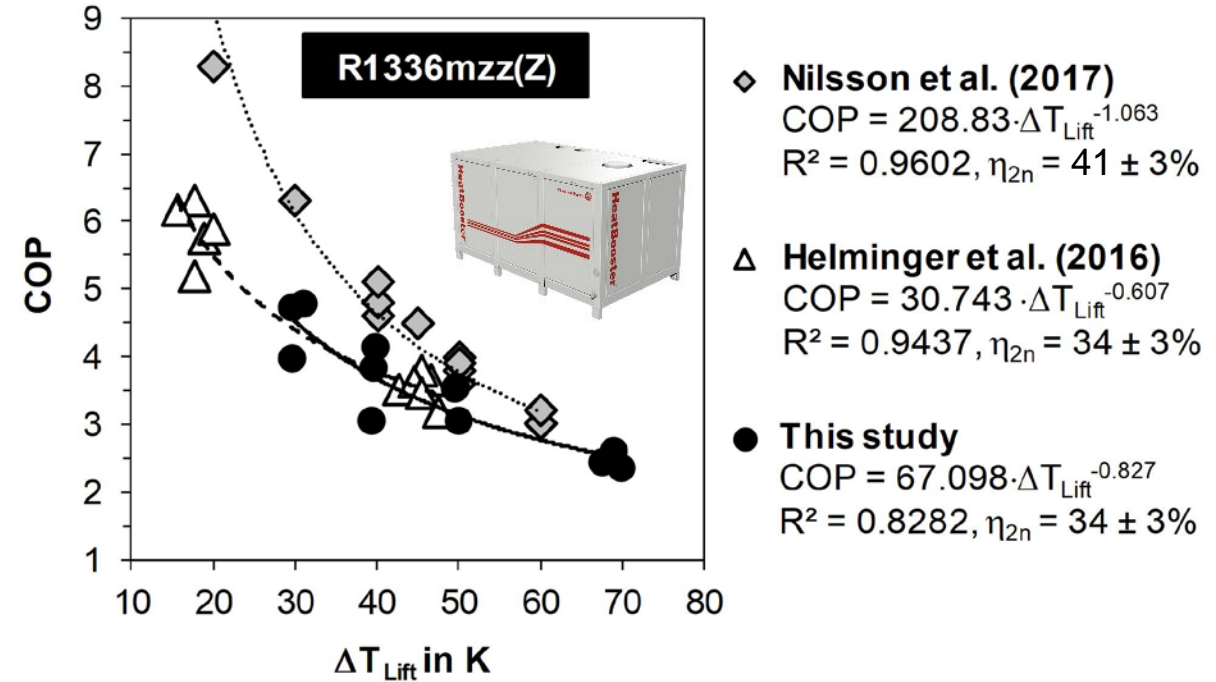
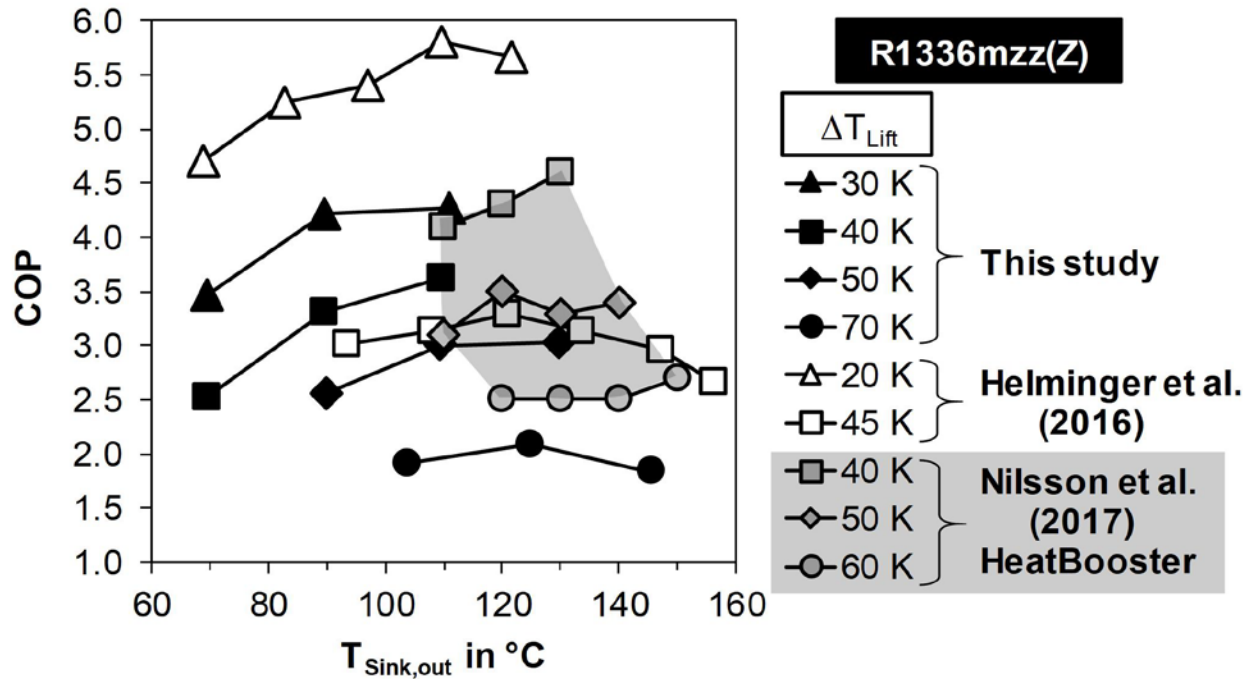
- R1233zd(E) and R1224yd(Z) provide higher COPs compared to R1336mzz(Z) up to about 110°C
- The smaller heating capacity of R1336mzz(Z) leads to more significant relative heat losses
- R1336mzz(Z) achieves potentially higher condensing temperatures ($T_{\text{crit}} = 171.3^\circ\text{C}$)
- R1224yd(Z) efficiency drops at 150 °C (close to the critical temperature of 155.5°C)

Comparison of heating capacity and COP fit curves with 2nd Law efficiencies



- R1233zd(E): heating capacity of 5.8 kW at Ref and approx. 10 kW at W80/W110
- R1336mzz(Z): maximum heating capacity of 7.8 kW at W80/W111
- R1224yd(Z): heating capacity on average 9% higher than R1233zd(E)

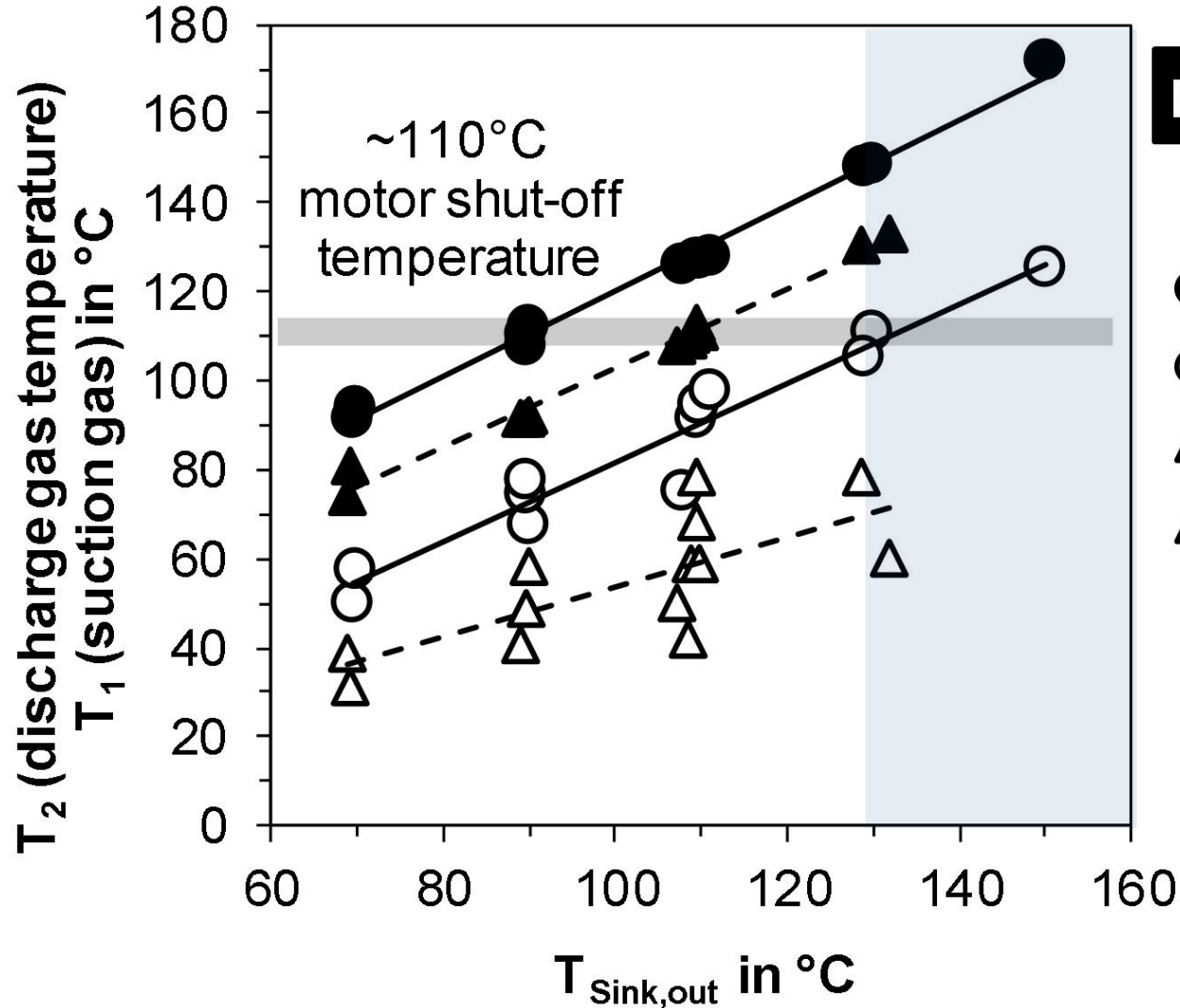
Comparison of COP data from Helminger et al. (2016) and Nilsson et al. (2017)



- Helminger et al. (2016):
 - 156.3°C and COP of almost 2.7 at 45 K ΔT_{Lift}
 - 2nd Law efficiency of $34 \pm 3\%$ similar to this study

- Nilsson et al. (2017):
 - Commercial HeatBooster technology achieves higher COP and 2nd Law efficiency of $41 \pm 3\%$
 - COP of 2.5 at 60 K ΔT_{lift}
 - With larger heating power the relative heat losses are getting smaller

Discharge and suction gas temperatures for tests with R1336mzz(Z)



R1336mzz(Z)

- T_2 } with IHX
- T_1 }
- ▲ T_2
- △ T_1



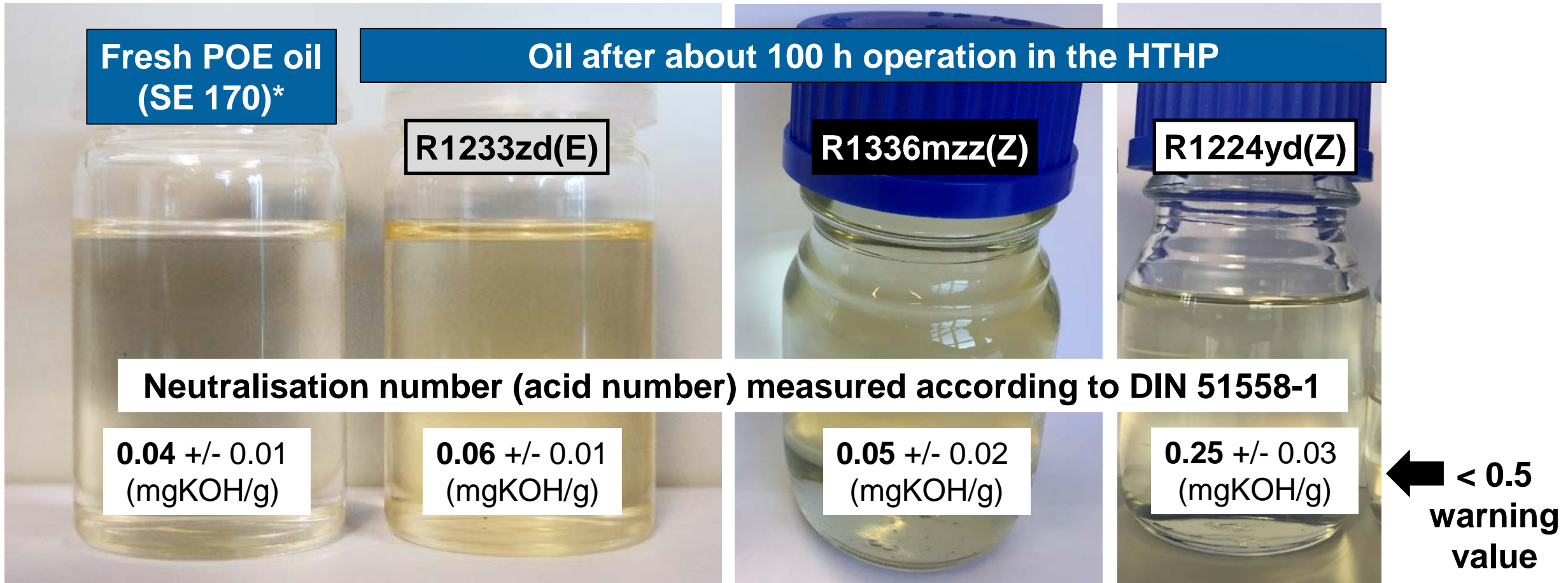
Basic cycle:

- Suction temperature (ΔT_1) well below the motor shut-off temperature of approx. 110 °C (grey line)

Cycle with IHX:

- Suction temperature (○ T_1) exceeded the motor limit temperature at a heat sink outlet temperature of about 130 °C and higher.

Negligible oil degradation after about 100 h operation in the HTHP



* 0.03 mgKOH/g according to product information from FUCHS



RENISO TRITON SE 170 – synthetic oil based on polyolester (POE) suitable for HFO refrigerants (complete miscibility with R1233zd(E) and R1336mzz(Z) between +100°C and -40°C)

Conclusions

- **R1336mzz(Z), R1233zd(E) and R1224yd(Z) successfully tested in single -stage lab-scale HTHP with IHX cycle and up to 10 kW**
- **Operation of demonstrated at 30 to 80°C heat source and 70 to 150°C heat sink temperatures (30 to 70 K temperature lifts) for possible application of waste heat recovery, steam generation or drying**
- **At W60/W110 COPs of 3.2, 3.1 and 3.0 for R1224yd(Z), R1233zd(E) and R1336mzz(Z) were measured**
- **Up to about 110 °C, R1224yd(Z) and R1233zd(E) slightly higher COP than R1336mzz(Z) due to higher heating capacities and lower relative heat losses at the same temperature conditions**
- **At 150 °C R1233zd(E) and R1336mzz(Z) more efficient than R1224yd(Z) due to higher critical temperatures**
- **Integration of an IHX increased COP (+15 to 47%) and heating capacity significantly**
- **Negligible oil degradation after about 100 h operation in HTHP (acid numbers < 0.5 mgKOH/g level)**
- **Very low GWP, non-flammability, and negligible environmental impact (low TFA formation during atmospheric degradation) indicate a high potential for future use as refrigerant in HTHP applications and retrofit systems**

Acknowledgements

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Thank you for your attention



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